SIMPLY THE FACTS
On
Animal Waste
and
Water Monitoring
(For Landowners, Farmers, & General Public)

A compilation of fact sheets on water quality monitoring by:

USDA Natural Resources Conservation Service
University of California Cooperative Extension
AmeriCorps
Water Quality Fact Sheets

This packet contains a series of water quality fact sheets concerning animal waste in surface waters. The fact sheets were designed to provide a clear and simple description of water monitoring tests performed by the California Department of Fish and Game, and the Regional Water Quality Control Board.

If you would like additional copies, please contact the Natural Resources Conservation Service in Petaluma at (707)794-1242.

Topics Covered

- Total Ammonia
- Unionized Ammonia
- pH
- Conductivity
- DO (Dissolved Oxygen)
- Monitoring Sites
- BOD (Biochemical Oxygen Demand)
- Water Quality Variables: Overview
- Ammonia Test Kit: Directions
- Water Testing: Techniques
- Ammonia, pH, Temperature: Tables

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SIMPLY THE FACTS ON
Total Ammonia

DEFINITION  Total ammonia is a measure of unionized and ionized ammonia in a sample of water. The degree of its toxicity changes with pH, temperature, and salinity. Organic nitrogen comes from all organic material such as manure, plants, dead animals, and fertilizers. Ammonia is a by-product of the decomposition of organic nitrogen. Total Ammonia (NH₃+NH₄⁺) is the amount monitored in creeks and ponds. It is from this measurement that the concentration of toxic unionized ammonia (NH₃) is calculated. Unionized ammonia (NH₃) is the toxic form that kills fish and aquatic life. (See unionized ammonia fact sheet) Monitoring is measured in parts per million (ppm) or mg/l.

WATER MONITORING  Total Ammonia

GOOD
Below 1 ppm
• Management practices are sufficient

CAUTION
1 ppm <= 5 ppm
• Nutrients entering surface water
• Evaluate waste management system

DANGER
Above 5 ppm
• Immediate action necessary!

AMMONIA SOURCES  Runoff from enclosed confined facilities (i.e., loafing barn). Runoff from silage storage areas. Runoff from open confined areas (i.e., corrals, feedlots). Runoff from manure storage areas. Facilities waste water. Runoff from over-fertilization. Runoff from overstocked pastures. Animals in the creek.
SIMPLY THE FACTS on

Unionized Ammonia

DEFINITION • Unionized ammonia (NH₃) is sometimes referred to as toxic ammonia and is the form which is harmful to fish and aquatic life. The toxicity level of unionized ammonia is directly affected by pH and temperature. The higher the pH and temperature of the water, the higher the proportion of total ammonia that exist in the toxic form. Unionized ammonia is what the California Department of Fish and Game and the Regional Water Quality Control Board regulate under the federal Clean Water Act. Measurements are derived from calculations using the pH, temperature, and total ammonia (NH₃ + NH₄⁺) content in the water. Monitoring is measured in parts per million (ppm) or mg/l.

WATER MONITORING • Unionized Ammonia (NH₃)

GOOD

0 - 0.025 ppm
• Management practices are sufficient

DANGER

0.025 ppm & Above
• Nutrients entering surface water
• Immediate action necessary!

POSSIBLE SOURCES • Unvegetated streambanks have higher water temperatures. Runoff from enclosed confined facilities (i.e., loafing barns). Runoff from silage storage areas. Runoff from open confined areas (i.e., corrals, feedlots). Runoff from manure storage areas. Facilities waste water. Runoff from over-fertilization. Runoff from overstocked pastures. Animals in the creek.
DEFINITION  pH is the measure of acid or alkaline characteristics and is measured on a scale of 0-14. A pH of 7.0 is neutral and ideal for water conditions. Readings that fall below 7.0 identify acidic conditions while readings above 7.0 identify alkaline conditions. The pH directly affects the amount of unionized ammonia in water. An increase in pH values above 7.0 combined with high water temperatures will result in higher levels of unionized ammonia which are deadly to fish. Shifts in pH levels can be contributed to a number of agricultural practices such as animal waste and silage, and should be kept away from streams. Changes in pH can also result from natural conditions such as algal growth.

WATER MONITORING  •  pH

GOOD
6.5 - 8.5
• Neutral
• Ideal water condition

CAUTION
5.0 <-> 6.5 and 8.5 <-> 10
• Readings closest to 7.0 are ideal
• Evaluate waste management

DANGER
0 <-> 5.0 and 10 <-> 14
• Acid and alkaline conditions
• Identify sources of pH change
• Immediate action necessary!

POSSIBLE SOURCES  Runoff from enclosed confined facilities (i.e., loafing barn). Runoff from silage storage areas. Runoff from open confined areas (i.e., corrals, feedlots). Runoff from manure storage areas. Facilities waste water. Runoff from over-fertilization. Runoff from overstocked pastures. Animals in the creek.
SIMPLY THE FACTS on DO

DEFINITION  Dissolved Oxygen (DO) is the concentration of oxygen dissolved in water. All fish and aquatic life must have adequate amounts of DO in the water at all times to survive. Problems occur when organic material such as animal waste enters surface water. Micro-organisms in the water use the organic material for food and consume DO in the process. When surface water with warm temperatures are mixed with organic material, a growth explosion of micro-organisms will occur, thus reducing DO to a level that suffocates fish and other aquatic life. Monitoring is measured in parts per million (ppm) or mg/l.

WATER MONITORING  •  DO

7 ppm <-> 10 ppm
• Good DO levels for aquatic life

5 ppm <-> 7 ppm
• DO levels moderate
• Evaluate waste management system
• Organic material may be entering surface water

0 ppm <-> 5 ppm
• Inadequate DO levels for most aquatic life
• Identify sources of organic material runoff
• Immediate action necessary!

LOW DO SOURCES  Unvegetated streambanks have higher water temperatures. Runoff from enclosed confined facilities (i.e., loafing barn). Runoff from silage storage areas. Runoff from open confined areas (i.e., corrals, feedlots). Runoff from manure storage areas. Facilities wastewater. Runoff from over-fertilization. Stagnate non-moving water. Runoff from overstocked pastures. Animals in the creek.

AmeriCorps &
United States Department of Agriculture
Natural Resources Conservation Service
(formerly Soil Conservation Service)

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For Water Quality Information contact:
Natural Resources Conservation Service
Regional Water Quality Control Board
California Department of Fish & Game
UC Cooperative Extension
SIMPLY THE FACTS on

BOD

DEFINITION  Biochemical oxygen demand (BOD) is an indirect way of measuring the oxygen required for the decomposition of materials such as animal waste in waters. A BOD test may be necessary when oxygen hungry bacteria contained in animal waste discharges into streams or ponds through runoff. This reduces the amount of oxygen available for fish. In reality it is a measurement of the ability of bacteria to grow and reproduce in animal waste over a 5 day period. Monitoring is measured in parts per million (ppm) or mg/l.

WATER MONITORING • BOD

GOOD

Below 10 ppm
• Management practices are acceptable

CAUTION

10 – 30 ppm
• Nutrients entering stream
• Evaluate waste management

DANGER

Above 30 ppm
• Immediate action necessary!

POSSIBLE SOURCES  Runoff from enclosed confined facilities (i.e., loafing barn). Runoff from silage storage areas. Runoff from open confined areas (i.e., corrals, feedlots). Runoff from manure storage areas. Facilities wastewater. Runoff from over-fertilization. Runoff from overstocked pastures. Animals in creeks. Decomposing leaves and grass debris in water.
DEFINITION  Electrical conductivity is one way to determine the salinity in soil and water. Salinity is the concentration of dissolved salts found in soil and water. An electric conductivity meter is used to determine salt content and is recorded in umhos/cm. High soil salinity interferes with plant water uptake resulting in reduced plant growth and germination. In excessive amounts, salts running off into nearby waters such as streams, ponds, and lakes can become toxic to freshwater aquatic plants and fish. Animal wastes as well as some agricultural products may have a high salt content and can be a problem when over-applied to the land. One should expect to find high salinity readings present in streams with tidal influences. If your property is located near an estuary, contact the Regional Water Quality Control Board to determine ideal salinity levels.

WATER MONITORING • Conductivity

![Good](image)
Below 500 umhos/cm
- Management practices are sufficient

![Caution](image)
Between 500 - 1000 umhos/cm
- Evaluate waste management system

![Danger](image)
Above 1000 umhos/cm
- Identify possible sources
- Immediate action necessary!

SALINITY SOURCES  Poor subsurface drainage. Irrigated crops. Runoff from enclosed confined facilities (i.e., loafing barn). Runoff from silage storage areas. Runoff from open confined areas (i.e., coral, feedlots). Runoff from manure storage areas. Facilities wastewater. Runoff from over-fertilization. Runoff from overstocked pastures. Animals in the creek. Tidal influence.
SIMPLY THE FACTS on
Monitoring Sites

To select water monitoring sites for a farm or ranch:

1. Obtain current aerial photography or a map of the property. Label the map in a numeric system to denote the best locations for possible monitoring sites (example below). Start by choosing sites on property lines (i.e., sites 1, 2A, 3A, 4). This will determine the water quality entering and leaving your property.

2. Pick sites located at the downstream end of tributaries to determine the water quality entering the main channel of a creek (i.e., sites 2 and 3). This way, you can identify possible sources of nutrient loading simply by backtracking upstream from those sites (to sites 2A and 3A). While backtracking, if you receive a clean/non-polluted sample, you can pinpoint the pollution source.

3. Once you have identified a source of pollution, keep a record of your results and note problem sites. Record any observations concerning water quality and riparian habitat. These observations include date, time, weather conditions, water color, odor (if any), vegetation, the presence of wildlife, and general conditions. Also record your water sampling data such as temperature, pH, ammonia, flow, turbidity, conductivity, DO, etc.

4. The best time to sample is in the morning when the water is coolest. It is also best to sample in the rain or just after a rain to locate problem areas. Consistency in sampling is important to evaluate results over a period of time and to make corrections as needed.
WATER MONITORING EXAMPLES

- A high nutrient reading was found at monitoring stations #1 and #1A. All other stations were sampled with low readings.
  - Conclusion: Nutrient source most likely is runoff from irrigated crop.

- A high turbidity (suspended sediment) level is recorded entering the property at station #4. At station #1 leaving the property a low turbidity level is recorded.
  - Conclusion: Sediment is being deposited between Station #1 and #4 in the stream channel.

- Station #2A and #2 have high nutrient levels recorded. A moderate nutrient reading was recorded at station #1A. Station #1B has a low nutrient level.
  - Conclusion: Tributary #2 has a high nutrient level entering the creek. Reading at station #1A has been diluted but a problem has still been diagnosed.
Thank You

The Natural Resources Conservation Service would like to say thank you to local farmers, landowners, contributing agencies, and AmeriCorps members for their participation in local water quality efforts.

"The CHP doesn't accept the excuse that you didn't know how fast you were going. The water cops and the public won't accept any excuses either. A producers' best protection is full knowledge of their individual situation. Then action can be taken or not taken as needed - and most importantly on your own terms and schedule. Get busted, and someone else will be setting the rules. Testing is easy and cheap."

Paul Martin, Dairy Farmer
Director, Southern Sonoma County
Resource Conservation District

"We have worked hard on the water quality committee organizing and using the tools that are available to us. However, there are still improvements to be made if we are to stay in business as dairy farmers. With hard work hopefully we can improve conditions for the dairy farmers while improving water quality, and protecting fish and wildlife habitat."

Joey Mendoza, Dairy Farmer
Marin-Sonoma Animal Waste Committee

"Monitoring water quality provides very valuable feedback. Operating without it is like driving with one's eyes closed."

Rick Bennett, Environmental Science Advisor
University of California Cooperative Extension

"The continued survival of local fish and wildlife resources is dependent upon our success in limiting the release of deleterious constituents from animal wastes into local creeks and streams. Death of aquatic life takes just a few minutes, but the effects are long term. Everyone needs to do their part."

Mike Rugg, Water Quality Biologist
California Dept. of Fish & Game

"We prefer to implement effective water quality management in partnership with agricultural landowners through voluntary compliance with water quality laws. Implementation of a voluntary monitoring program will provide an assessment of water quality, help identify any problems, and will measure the success of implemented corrective actions. The health and welfare of people and food producing animals is dependent on good water quality. Good water quality is also critical for the aquatic life, birds, and mammals that depend on our waterways for habitat and food. Please join us and help improve and protect water quality in your watershed. Your commitment to this partnership is appreciated."

Dennis Salisbury, Environmental Specialist
Regional Water Quality Control Board