Upper Newport Bay
Water Quality Model Development

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EXECUTIVE SUMMARY

Newport Bay has been listed as an impaired water body due to eutrophication. A TMDL program has resulted in reduced nutrient inflows in the Bay, but macroalgal blooms continue to impair beneficial uses. The TMDL implementation plan requires determination of new numeric nutrient water quality objectives for freshwater tributaries to Newport Bay.

The Regional Board has funded Resource Management Associates (RMA) to develop water quality model of Newport Bay that includes macroalgae. The new model is an expansion of a previous version of the model that successfully simulated the occurrence of macroalgal blooms, but could not precisely predict yearly variations in the blooms. An interesting study by Southern California Coastal Water Research Project (SCCWRP) suggests that sediment fluxes play an important role as an internal loading of nutrients in Newport Bay. It was hypothesized that the magnitude of the macroalgal bloom in any given year is strongly dependent on the amount of sediment entering the Bay during the previous wet season. Based on the SCCWRP study and a thorough review of literature on the subjects of macroalgae, sediment processes and Newport Bay in general, the water quality model has been updated to include sediment nutrient fluxes. The updated model has been calibrated and validated using new data sets from an annual monitoring program performed by the Regional Board from 2004 through 2006 so that it can be run in a predictive mode to evaluate the impacts of alternative nutrient load levels entering the Bay. These simulations will be the main tool for selecting appropriate numeric water quality objectives.

Conceptual models for Newport Bay describe a linkage between macroalgal blooms and nutrient cycles. The main constituents of the water quality model include macroalgae, nitrate, ammonia, phosphate, dissolved oxygen, and deposited particulate organic matter (POM). DO, POM, NH3, NO3 and PO4 are considered in the sediment flux model. Concentrations of DO, NO3, NH3 and PO4 in the sediments affect water column concentrations. A separate sediment transport simulation is run to provide mass and distribution of deposited sediment to the sediment flux model.

The model is calibrated for January through December 2004. The validation period is from January 2005 through December 2006. These periods include a year with a dryer winter and smaller blooms later in the year (2004), and a year with large winter flows and subsequent larger macroalgal blooms later in the year (2004 – 2005). The validation is extended into 2006 to cover another dryer year.

Model simulations were run with nutrients simulated as conservative constituents to help determine the relative importance of the tidal and creek inflow boundary contributions to total nutrient concentrations in the Bay. Similarly, simulations were run with biological and chemical reactions but no sediment flux and compared with the calibrated model. Each of these simulations helped discern the importance of the different processes and sources.
Residence time simulations were run for a two-month summer period to determine the range of residence times that occur in the Bay over the spring-neap tidal cycle. Peak residence time is around 12 days, occurring in Rhine Channel.

The calibrated model was run in a predictive mode to evaluate the impacts of alternative total nitrogen load levels entering the Bay. Six alternative nitrogen loading levels were analyzed with different total nitrogen limits during storm and low flow periods. The fraction of particulate nitrogen in the suspended sediment was set to reduced levels corresponding to the storm flow nitrogen limits. Alternative simulation results indicate that reductions in total nitrogen in storm flows and fraction of particulate nitrogen in the sediment have a greater impact on reducing macroalgae blooms than reduction in total nitrogen inputs during low flow periods. Sediment flux appears to be one of the most important sources of nutrients for macroalgae growth.
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1. Introduction

1.1 Background

Newport Bay has been listed as an impaired water body due to eutrophication, and a total maximum daily load (TMDL) for nutrients was adopted by the Santa Ana Regional Water Quality Control Board (Regional Board) in 1998. The TMDL program has resulted in reduced nutrient inflows in the Bay, but macroalgal blooms continue to impair beneficial uses. The TMDL implementation plan requires determination of new numeric nutrient water quality objectives for freshwater tributaries to Newport Bay. Calibration of these new objectives is a complex scientific problem; the U.S. EPA is still in the process of formulating methodologies for determining appropriate numeric objectives.

For a better understanding of the macroalgal blooms, the Regional Board has funded development of a water quality simulation model for macroalgae in the Newport Bay. Development of the Newport Bay model was originally funded by the U.S. Army Corps of Engineers for evaluation of alternative dredging options for Newport Bay (RMA, 19979; 1998a; 1998b; 199). The Regional Board subsequently funded expansion of the model to simulate macroalgae and other water quality constituents (RMA, 2003).

Previous versions of the model have successfully simulated the occurrence of macroalgal blooms, however, the model could not precisely predict yearly variations in the blooms. A study by Southern California Coastal Water Research Project (SCCWRP) (Sutula et al., 2006) suggests that sediment fluxes play an important role as an internal loading of nutrients in Newport Bay. It was hypothesized that the magnitude of the macroalgal bloom in any given year is strongly dependent on the amount of sediment entering the Bay during the previous wet season. Based on SCCWRP study, a thorough literature review and monitoring data collected by the County of Orange from 2001 to 2007, the model has been updated with sediment nutrient fluxes, and calibrated and validated using the new data sets from the annual monitoring program from 2004 to 2006.

The water quality model has thus been updated to include sediment fluxes, re-calibrated using the new data sets and run in a predictive mode to evaluate the impacts of alternative nutrient load
levels entering the Bay. These simulations are intended to assist the Regional Board in selecting appropriate numeric water quality objectives for freshwater creeks that drain into the Bay.

1.2 Objectives

The objective of the current Upper Newport Bay Water Quality Model Development project is to update and calibrate the water quality model and perform simulations to evaluate alternative loading strategies in support the short term and long term management of Newport Bay. The project consists of the following:

- Modification of the water quality model, including the addition of a sediment flux model;
- Re-calibration and validation of the water quality model over the three-year period of 2004 through 2006; and
- Evaluation of reduced total nitrogen loading alternatives.
2. Literature review

2.1 Characteristics of macroalgal blooms

Cultural eutrophication of coastal ecosystems has increased largely due to augmented supplies of nutrients (i.e., nitrogen (N) and phosphorus (P)) from the watershed, of which land uses are dominated by agricultural activity and urbanization (Hauxwell and Valiela, 2004; Kamer et al., 2001).

Nitrogen tends to be the more limiting nutrient controlling eutrophication of coastal waters because:
- nitrogen fixation generally does not occur in estuaries with salinities > 10 to 12 even when they are strongly nitrogen limited;
- denitrification tends to be one of the major sinks (although this may be slightly less important in ecosystems with shorter water residence times, such as Newport Bay); and
- phosphorus availability in estuaries is greater relative to other water bodies due to desorption from suspended matters under high salinity conditions (Howarth and Marino, 2006).

Thus, as inputs of nitrogen to a coastal system such as Newport Bay increase, the growth limitation is reduced. The increases of nutrient inputs, especially nitrogen, have changed the composition and abundance of the primary producers present in coastal systems. Primary producers in coastal ecosystems are typically made up of various forms of microalgae (benthic, epiphytic and pelagic), and macroalgae (ephemeral and persistent) and rooted macrophytes (Sand-Jensen and Nielsen, 2004; Pedersen et al., 2004). In particular, periodic and massive macroalgal blooms, such as *Ulva*, *Cladophora*, *Enteromorpha* and *Chaetomorpha*, have become common phenomena in temperate, relatively nutrient-rich waters where the bottom is within the photic zone (Naldi and Viaroli, 2002; Valiela et al., 1997). The three most common species of macroalgae in Newport Bay are *Ulva*, *Enteromorpha* and *Ceramium* (see Figure 2-1).

Even though both macroalgal and microalgal blooms may have intensive consequences, macroalgal blooms differ from microalgal blooms in three major ways (Valiela et al., 1997). Macroalgal blooms:
- lack direct chemical toxicity;
- have a broader range of ecological effects (i.e., thoroughly altering function and structure of affected ecosystems); and
- have a longer life.
Macroalgal blooms show strong seasonal patterns which are often related to large inputs of nutrients (Kamer et al., 2001). In the case of green macroalgae, such as *Enteromorpha* and *Ulva*, they can store both nitrogen and phosphorus and their growth rates are considered to be dependent on internal nutrient concentrations, while their uptake rates depend on the difference between internal nutrient concentration in macroalgae cells and external nutrient concentration in the water (Martins et al., 2001). Some of macroalgal species (e.g., *Enteromorpha intestinalis*) have preference of ammonium to nitrate in the water column for nitrogen uptake (Cohen and Fong, 2004).

In the case of *Enteromorpha*, in spite of their ability to fix to substrata on the sediment, it cannot completely resist the water flow. Strong tidal currents erode the attachment of *Enteromorpha* or tear away their fronds once they reach a critical length. This detachment was found in Langstone Harbour (U.K.), where the maximum current velocity is about 1.22 m s⁻¹ (Martins et al., 2001).

![Ulva](image1.png) ![Enteromorpha](image2.png) ![Ceramium](image3.png)

Figure 2-1. Three common species of macroalgae appeared in Newport Bay.

### 2.2 Characteristics of macroalgae modeling

The following controls on macroalgal blooms are considerations for the model formulation.

- **Temperature** (Martins and Marques, 2002; Valiela et al., 1997).
- **Light** (Martins and Marques, 2002; Valiela et al., 1997). The formation of layered biomass mats has the potential to affect photosynthesis and respiration via various light attenuations (Brush and Nixon, 2003).
- **Limiting nutrients** (nitrogen, phosphorus) (Kamer et al., 2001; Martins and Marques, 2002; Valiela et al., 1997). Increased nutrient supply increases nutrient uptake rates,
tissue nutrient contents, and growth rates of macroalgae (Valiela et al., 1997). Macroalgae in the Bay have a relatively low tissue N:P (molar) ratio (e.g., 16.4-28.1 for Enteromorpha intestinalis; 21.8-40.1 for Ceramium spp.; 19.8-30.0 for Ulva expansa) which may indicate nitrogen limitation (Kamer et al, 2001). Therefore, nitrogen is likely the most limiting nutrient in Newport Bay, and phosphorus is the secondary (Kamer and Schiff, 2002).

Aside from creek inflows, possible sources for nutrients in the Bay include the following.

- **Upwelling:** quite large amounts of nitrate (approximately 0.2 mg L⁻¹) can be present in ocean water since the west coast of the Pacific Ocean is an upwelling zone (AHA, 1997). The ocean current, carrying cold water from northern latitudes, flows along the shore from Washington-Oregon border to southern California. Seasonal winds, combined with the Coriolis effect, drive surface waters offshore beginning in the spring. The driven surface waters from the coast are replaced by denser, cold water below carrying dissolved nutrients (http://ceres.ca.gov/ceres/calweb/coastal/waters.html). This process, known as upwelling, generally continues until September.

- **Nutrients released by sediments:** organic materials delivered from inflows are decomposed in the sediments and the remineralized nutrients are released from the sediments.

- **Salinity** (Martins and Marques, 2002; Martins et al., 1999). Martins et al., (1999) showed that the growth of Enteromorpha intestinalis varies along a bell-shaped curve with salinity and that the optimum salinity range for growth is 18-22 ppt. E. intestinalis is more affected by lower than by higher salinity.

- **Grazing rate** (Valiela et al., 1997).

- **Respiration and mortality rate** (Trancoso et al., 2005).

- **Detachment stress** (Martins and Marques, 2002; Trancoso et al., 2005).

Although macroalgae and microalgae (e.g., phytoplankton) are subject to the same growth requirements (temperature, light and nutrients) and the same basic processes, major differences between the two (Trancoso et al., 2005) must be taken into account in the model:

- **Macroalgae** are associated with the bottom substrate and are expressed in terms of areal densities (kg m⁻²) rather than volumetric densities or concentrations (mg L⁻¹).

- **Macroalgae** are not dominantly subject to hydrodynamic transport.

- **Macroalgae** have no settling losses, but instead they have additional losses by sloughing and scouring from the bottom substrate where bottom shear stress is high.

- **Macroalgal growth** varies areally due to bed elevation because of light limitation and tidal stage. Light limitation reduces growth at bed elevations below the optimal elevation, and prevents colonization altogether at the lowest bed elevations. At higher bed elevation,
macroalgae can become emerged at low tide. Emerged macroalgae are considered to be dormant, reducing the overall growth rate, and extended emergence results in mortality. The productivity yield after reimmersion is not affected unless emergence results in mortality.
3. Conceptual models for macroalgal blooms in Newport Bay

The conceptual model for Upper Newport Bay describes the linkages between nutrient inputs and other parameters (temperature, light, salinity, etc.) and macroalgal blooms. The major constituents of the conceptual model include macroalgae, nitrogen (nitrate, ammonia, and particulate organic nitrogen), phosphorus (phosphate), dissolved oxygen, salinity, temperature and particulate organic matter, as shown in Figure 3-1.

These constituents are inter-related through biological, chemical, and physical processes in the Bay. The conceptual model is used to build the water quality model, which consists of a set of governing equations.

Table 3-2 lists the variables and symbols used in the water quality model. The coefficients, constants, and forcing parameters of the water quality model are shown in Table 3-3 and the governing equations are shown in Table 3-4.
Figure 3-1. Schematic of the ecological model (with major nutrient interactions) of Newport Bay. Boxes indicate the constituents and arrows indicate biological, chemical and physical processes.
3.1 Macroalgae

The macroalgal community of Newport Bay is dominated by three main species including *Enteromorpha intestinalis*, *Ulva expansa* and filamentous red *Ceramium spp.* (Kamer et al., 2001). In the conceptual model, the three species are not modeled specifically, rather rates and coefficients are set to best characterize the overall macroalgae growth dynamics in the Bay. Macroalgal dynamics are governed in the model by the following processes: growth, respiration (and excretion), sloughing, and mortality, as shown in Figure 3-2. The concentration of macroalgae \( M \) changes as follows:

\[
\frac{dM}{dt} = (\mu - r - s - m) \times M
\]  

(3.1)

where \( \mu \) is growth, \( r \) is respiration, \( s \) denotes sloughing, and \( m \) is mortality.

Figure 3-2. Conceptual model for macroalgae in Newport Bay.
3.1.1 Growth

The growth rate of macroalgae is described as a function of maximum growth rate at the optimum temperature ($\mu_{\text{max}}$), temperature ($f(T)$), light ($f(L)$), salinity ($f(S)$), space ($f(SF)$), water depth ($f(D)$) and nutrients ($f(QN)$, and $f(P)$) where $QN$ indicates internal nitrogen quota of macroalgae.

$$
\mu = \mu_{\text{max}} \times f(T) \times f(L) \times f(S) \times f(SF) \times f(D) \times \min(f(QN), f(P))
$$

(3.2)

(1) Temperature function

The temperature function is computed as:

$$
f(T) = \exp\left(-2.3 \times \left(\frac{T - T_{\text{opt}}}{T_{x} - T_{\text{opt}}}\right)^2\right)
$$

where $T_{\text{opt}}$ is the optimum temperature at which growth rate is maximum

$$
\text{if } T < T_{\text{opt}} \quad T_{x} = T_{\text{min}}
$$

$$
\text{if } T > T_{\text{opt}} \quad T_{x} = T_{\text{max}}
$$

where $T_{\text{min}}$ is lower temperature limit below which growth ceases and $T_{\text{max}}$ is upper temperature limit above which growth ceases (Martins et al., 2007; Martins and Marques, 2002).

(2) Salinity function

The salinity function is computed as:

$$
f(S) = 1 - \left(\frac{S - S_{\text{opt}}}{S_{x} - S_{\text{opt}}}\right)^m \quad \text{for } S \geq 5
$$

$$
f(S) = \left(\frac{S - S_{\text{min}}}{S_{\text{opt}} - S_{\text{min}}}\right)^m \quad \text{for } S < 5
$$

(3.4)

where $S_{\text{opt}}$ is the optimum salinity at which growth rate is maximum

$$
\text{if } S < S_{\text{opt}} \quad S_{x} = S_{\text{min}} \text{ and } m = 2.5
$$

$$
\text{if } S \geq S_{\text{opt}} \quad S_{x} = S_{\text{max}} \text{ and } m = 2
$$
where $S_{\text{min}}$ is lower salinity limit below which growth ceases and $S_{\text{max}}$ is upper salinity limit above which growth ceases (Martins et al., 2007; Martins and Marques, 2002).

(3) Light function

There are three options for the light function in the model:

1) Half Saturation

$$f(I) = \frac{I_z}{k_{ls} + I_z}$$

(3.5)

2) Smith function

$$f(I) = \frac{I_z}{\sqrt{k_{ls}^2 + I_z^2}}$$

(3.6)

3) Steele formula

$$f(I) = \frac{I_z}{I_o} \exp\left(\frac{I_z}{I_o}\right)$$

(3.7)

where $I_z$ is light intensity at depth $z$, $k_{ls}$ is a half saturation constant for light; $I_o$ is optimal light level (Bowie et al., 1985). In this study, the Steele formula of the light function was used for the simulation of macroalgae. The Steele light function includes macroalgal self shading and was selected because it resulted in the best fit to the macroalgal data among three formulas. The attenuation of light in natural waters can be quantified by the Beer-Lambert law,

$$i_z = I_o \exp (-\lambda z)$$

(3.8)

where $\lambda$ is the light attenuation coefficient and $I_o$ is surface light intensity (energy). For the macroalgae modeling, the light intensity was calculated at the bottom of the water column. The light extinction coefficient is coupled to the macroalgal concentration using the non-linear Riley equation (Bowie et al., 1985):

$$\lambda = \lambda_0 + \lambda_1 M$$

(3.9)

where $\lambda_0$ is non-algal portion of the light extinction coefficient, $(m^{-1})$ and $\lambda_1$ is linear macroalgal self-shading coefficient, $(kg m^{-3})$. 

11
(4) Space function
The macroalgal space limitation factor is defined as follows:

\[ f(SF) = 1 - \frac{H}{k_{sp}} \]  

(3.10)

where \( k_{sp} \) is the space limitation half-saturation constant for growth limitation (RMA, 2003).

(5) Depth function
The macroalgal depth limitation factor is expressed as:

\[
\begin{align*}
  f(D) &= 1 & \text{if } D > k_D \\
  f(D) &= 0 & \text{if } D < k_D
\end{align*}
\]  

(3.11)

where \( D \) is the water depth and \( k_D \) is the critical water depth for growth limitation (RMA, 2003). The critical depth is defined as shallow water (e.g., water depth at low tide).

(6) Nutrients function
The internal nitrogen limitation factor of macroalgae is expressed as:

\[ f(QN) = \frac{QN - QN_{\text{min}}}{k_q + QN - QN_{\text{min}}} \]  

(3.12)

where \( QN = \frac{IN}{M} \), where \( IN \) is internal nitrogen concentration of macroalgae; \( QN_{\text{min}} \) is the minimum internal nitrogen quota (i.e., minimum possible value for nitrogen concentration in macroalgal tissue) and \( k_q \) is the nitrogen half-saturation constant for growth limitation (Solidoro et al., 1997).

Chlorophyte macroalgae, like Ulva and Enteromorpha, have a lower internal phosphorus concentration than that of nitrogen. The range of internal phosphorus concentration is restricted only between 1 and 5 mg P (g dw)\(^{-1}\) (Solidoro et al., 1997). In addition, the critical concentration level for internal phosphorus, where the growth reaches its maximum efficiency, is close to the minimum quota value (Solidoro et al., 1997). Therefore, the growth of macroalgae is assumed to use externally available phosphorus rather than to store phosphorus in the biomass. The growth rate of macroalgae of Newport Bay is conceptualized in the model to be related to the external phosphorus.
concentration through the Monod kinetic. The phosphorus limitation factor of macroalgae is expressed as:

\[
f(P) = \frac{PO_4}{k_{PO_4} + PO_4}
\]  

(3.13)

where \( k_{PO_4} \) is phosphate half-saturation constant for growth limitation (Solidoro et al., 1997).

3.1.2 Respiration

Respiration is a function of temperature (Bowie et al., 1985) and is expressed as:

\[
r = r_{\text{max}} \times f(r)
\]  

(3.14)

where \( r_{\text{max}} \) is maximum respiration rate at the optimum temperature.

3.1.3 Sloughing

Macroalgae are sloughed by bottom currents, which transport them close to the bed (Flindt et al., 2007; Trancoso et al., 2005). Flindt et al. (2007) suggested a linear regression between sloughed biomass and current speed for four species of macroalgae. Their linear regressions for two kinds of macroalgae (i.e., \( Ulva \) sp. and \( Enteromorpha \) sp.) are presented in Table 3-1.

Table 3-1. Estimated sloughing thresholds and current speeds at which total sloughing is reached, calculated from linear regression between sloughed biomass and current speed for two species of macroalgae (source: Flindt et al., 2007).

<table>
<thead>
<tr>
<th>Species</th>
<th>Threshold (cm s(^{-1}))</th>
<th>Regression</th>
<th>( r^2 )</th>
<th>Total sloughing (cm s(^{-1}))</th>
</tr>
</thead>
<tbody>
<tr>
<td>( Ulva ) sp.</td>
<td>12</td>
<td>( y = 0.0143x-0.17 )</td>
<td>0.91</td>
<td>82</td>
</tr>
</tbody>
</table>
Enteromorpha sp. & 21 & $y = 0.0094x - 0.2$ & 0.86 & 128 \\

$y$ is a ratio of sloughing biomass to total biomass; $x$ is a current speed (cm s$^{-1}$)  
*No data is available for *Ceramium*.

In the model, when macroalgae are sloughed they are not transported; rather they are removed from the system at the time of sloughing (i.e., it is assumed that sloughed macroalgae exited the system to the ocean due to high water current speed, and did not settle and decay). The sloughing term in the model serves primarily to eliminate macroalgae in the deeper channels where the velocities are higher.

Sloughing rate for a time interval is expressed as follows:

$$ s = \frac{U - U_{\text{min}}}{U_{\text{max}} - U_{\text{min}}} \quad \text{if} \quad U \gg U_{\text{min}} $$  

(3.15)

where $U$ is current speed; $U_{\text{min}}$ is minimum current speed at which sloughing occurs and $U_{\text{max}}$ is maximum current speed at which sloughing of all the macroalgae occurs.

### 3.1.4 Mortality

In the conceptual model, mortality and algal decomposition are minimal when growth conditions are optimal, and maximum when conditions are severely limiting (Bowie et al., 1985). Macroalgal mortality is expressed as:

$$ m = m_{\text{max}} \times \left[1 - \left(f(S) \times f(U) \times \min\left(f(QN), f(P)\right)\right)\right] $$  

(3.16)

where $m_{\text{max}}$ is maximum mortality rate at the optimum temperature.

In the model, a certain portion (70 – 80%) of the dead macroalgae is assumed to be recycled as deposited on the bed where it decays and returns nutrients to the system. The percentage of dead macroalgae that was recycled was a calibrated parameter in the model.

It is important to note that macroalgae is modeled as a fixed constituent and is never transported. Therefore, recycled macroalgae enters the bed at the location where it grew.

### 3.2 Nitrogen

The conceptual model of nitrogen consists of nitrate, ammonia, and internal nitrogen (see Figure 3-3).
• Nitrate
Nitrate ($\text{NO}_3^-$) is provided with nitrification of ammonia ($\text{NH}_3^-$), but consumed by the uptake by macroalgae and denitrification of $\text{NO}_3^-$. Mass transfer of $\text{NO}_3^-$ between porewater and water column is in the model as well. The direction of mass transfer depends on the concentration gradient between the porewater and water column. Reaction rates of these processes are described as the first-order kinetics. The equation of nitrate dynamics is as follows:

$$\frac{d\text{NO}_3^-}{dt} = \left( k_{\text{nitr}} \times \text{NH}_3^- \times f_{\text{nitr}} - k_{\text{dinit}} \times \text{NO}_3^- \right) \times e^{T_{\text{1} - 20}} - \frac{V_{\text{NO}_3^-}}{D} - k_{\text{ENO}_3^2} (\text{NO}_3^- - \text{NO}_3^2_e)$$

$$V_{\text{NO}_3^-} = V_{\text{mNO}_3^-} \times \frac{\text{NO}_3^-}{k_{\text{ENO}_3^2} + \text{NO}_3^-} \times \frac{Q_{\text{N max}} - Q_N}{Q_{\text{N max}} - Q_{\text{N min}}}$$

where $k_{\text{nitr}}$ is the nitrification rate of $\text{NH}_3^-$; $k_{\text{dinit}}$ is the denitrification rate of $\text{NO}_3^-$; $T_{\text{1} - 20}$ is the non-dimensional temperature multiplier of reaction; $V_{\text{NO}_3^-}$ is nitrate uptake rate; $k_{\text{ENO}_3^2}$ is the mass transfer rate of NO3; $\text{NO}_3^2_e$ is the concentration of NO3 in the sediment; $Q_{\text{N max}}$ is maximum nitrate uptake rate; $k_{\text{ENO}_3^2}$ is half saturation constant for NO3; and $Q_{\text{N max}}$ and $Q_{\text{N min}}$ are the maximum and minimum internal nitrogen quotas for macroalgae, respectively.

The nitrification reaction depends on ammonium concentration and sufficient oxygen levels (greater than 1 to 2 mg L$^{-1}$). The nitrification inhibition factor, $f_{\text{nitr}}$, is expressed by a function of the concentration of dissolved oxygen (DO) as follows (Chapra, 1997):

$$f_{\text{nitr}} = 1 - \exp(-k_{\text{nitr}} \times \text{DO})$$

where $k_{\text{nitr}}$ denotes first-order nitrification inhibition coefficient ($\approx$ 0.6 L mg$^{-1}$). All other coefficients are previously defined.

• Ammonia
Ammonia ($\text{NH}_3^-$) is provided with respiration (or excretion) of macroalgae, and is consumed by nitrification and the uptake by macroalgae in the water column. $\text{NH}_3^-$ is also produced by mineralization of settled POM in the sediments. $\text{NH}_3^-$ in the sediments is
transferred to the water column corresponding difference of concentrations of two media. The mass transfer of \(NH_3\) is included in the model. The equation of ammonia dynamics is as follows:

\[ \frac{dNH_3}{dt} = r \times QN \times \frac{M}{D} - V_{NH_3} \times \frac{M}{D} - k_{NH_3} \times NH_3 \times f_{NH_3} \times q_{FNM} - k_{SNH_3}(NH_3^2 - NH_3^2) \]  

(3.20)

where \(V_{NH_3}\) is ammonia uptake rate; \(k_{NH_3}\) is the nitrification rate of \(NH_3\); \(k_{SNH_3}\) is the mass transfer rate of \(NH_3\); \(NH_3^2\) is the concentration of \(NH_3\) in the sediment; \(V_{NH_3}^{max}\) is maximum ammonia uptake rate; and \(k_{SNH_3}\) is half saturation constant for \(NH_3\). All other coefficients are previously defined.

- Internal nitrogen \(IN\) is provided with the uptake of \(NO_3\) and \(NH_3\) and consumed by respiration, sloughing and mortality of macroalgae.

\[ \frac{dIN}{dt} = (V_{NO_3} + V_{NH_3}) \times M - (r + s + m) \times IN \]  

(3.22)

where \(V_{NO_3}\) is the nitrate uptake rate by macroalgae and \(V_{NH_3}\) is the ammonia uptake rate. All other coefficients are previously defined.
Figure 3-3. Conceptual model for nitrogen in Newport Bay.

3.3 Phosphorus

In the conceptual model, orthophosphate ($P_{O4}$) (see Figure 3-4) is provided by respiration of macroalgae, and consumed by biological uptake and adsorption of $P_{O4}$ to $POM$. In the model, adsorption is included to account for accumulation of $P_{O4}$ on the surface of suspended solids in the water column which will be flushed out of system (adsorbed $P_{O4}$ does not contribute to the sediments in the model). $P_{O4}$ is also produced from settled $P_{O4}$ in the sediments which is included in the model as the mass transfer. There is no mineral phosphorus in the model. Rate constants of these processes are described as the first order kinetics. The equation describing orthophosphate dynamics is as follows:

$$\frac{dP_{O4}}{dt} = (r - \mu) \times \alpha_{P} \times \frac{M}{D} - K_{sp}P_{O4}(P_{O4} - P_{O4a}) - k_{ads} \times P_{O4} \times e^{-\gamma_{rav}}$$ \hspace{1cm} (3.23)

where $\alpha_{P}$ is a fraction of macroalgae that is phosphorus; $K_{sp}$ is the mass transfer rate of $P_{O4}$; $P_{O4}$ is concentration of $P_{O4}$ in the sediment; and $k_{ads}$ is a rate of adsorption onto suspended solids (day$^{-1}$). All other coefficients are previously defined.
Figure 3-4. Conceptual model for phosphorus in Newport Bay.

3.4 **Dissolved oxygen (DO)**

Dissolved oxygen (DO) is affected by nutrient cycles that generally demand oxygen. The conceptual model of dissolved oxygen in Newport Bay includes the processes of reaeration of O$_2$ gas (gas transfer), photosynthesis, respiration of macroalgae, nitrogenous biochemical oxygen demand (NBOD) for nitrification, and sediment oxygen demand (SOD) from nutrient-rich benthic sediment (see Figure 3-5). DO in the water column and sediments is exchanged via mass transfer. Reaction rates are described as first-order kinetics (Chapra, 1997).
DO concentration in the water column is described by the following equation:

\[
\frac{d[D_0]}{dt} = k_a \times \theta_{D_0} \times (O_2 - D_0) + (\mu - \tau) \times \alpha_c \times \alpha_c \times \frac{M}{D} - K_{sod} (D_0 - D_O) \times \theta^{T-20} - k_{mt} \times \theta_{D_0} \times NH_3 \times f_{mt} \times \theta^{T-20}
\]

where \( k_a \) is the reaeration rate; \( O_2 \) is the saturation concentration of oxygen at non-standard pressure, \( \theta_{D_0} \) is the ratio of mass of oxygen consumed per mass of carbon assimilated; \( \alpha_c \) is a fraction of macroalgae that is carbon; \( K_{sod} \) is the mass transfer rate of \( D_0 \) between sediment and water column; \( D_O \) is concentration of \( D_0 \) in the sediment; \( \theta_{D_0} \) is the ratio of mass of oxygen consumed per mass of nitrogen oxidized. All other terms have been previously defined.

- **Reaeration rate**
  
  Estuary gas transfer, \( k_a \), can be affected by both water and wind velocity. The water velocity effects are typically computed with the O’Connor-Bobbins formula (Chapra, 1997).
\[ k_a = \frac{D_{LQ} U_0}{H^{3/2}} \]  \hspace{1cm} (3.25)

where \( U_0 \) is mean tidal velocity over a complete tidal cycle; \( D_{LQ} \) is liquid diffusion coefficient. The diffusivity of oxygen in natural waters is approximately \( 2.09 \times 10^{-5} \) cm\(^2\) s\(^{-1}\).

The wind effects can be computed with any of the formulas developed for standing waters (Chapra, 1997) as follows:

\[ k_w = \frac{0.728 U_w^{0.6} - 0.317 U_w + 0.0372 U_w^2}{H} \]  \hspace{1cm} (3.26)

where \( U_w \) is wind speed measured 10 m and above the water surface (m s\(^{-1}\)). Combining the above two approaches for estuaries affected by both tidal velocities and wind gives the following:

\[ k_a = 3.93 \frac{\sqrt{U_0}}{H^{3/2}} + \frac{0.728 U_w^{0.6} - 0.317 U_w + 0.0372 U_w^2}{H} \]  \hspace{1cm} (3.27)

The formula both with water and wind velocity was applied for Newport Bay.

- **Saturation of dissolved oxygen (RMA, 2003)**
  Solubility of dissolved oxygen is dependent on temperature, salinity, and atmospheric pressure. The conceptual model limits the dependence to temperature, salinity and atmospheric pressure when temperature is a defined constituent.

The relationships take the following forms:

\[
\ln Q_m = -139.34410 + \frac{1.575701 \times 10^{-5}}{T_a} - \frac{6.642308 \times 10^{-7}}{T_a^2} + \frac{1.243800 \times 10^{-10}}{T_a^3} - \frac{8.621949 \times 10^{-11}}{T_a^4} - \frac{1}{1.60655 \times S_{wi} \times \left[ (3.1929 - 2) - \frac{1.9428 \times 10^{-1}}{T_a} + \frac{3.8673 \times 10^{-3}}{T_a^2} \right]} \]

where \( Q_m \) is equilibrium oxygen concentration at 1.00 atm (mg L\(^{-1}\)); \( T_a \) is water temperature (°K), within the range 0.0 to 40.0 °C; and \( S_{wi} \) is salinity (if active in the simulation).
\[ Q_e = Q_{eq}P \frac{(1 - P_{WV}/P)(1 - \varphi P)}{(1 - P_{WV})(1 - \varphi)} \]  

(3.29)

where \( Q_e \) is equilibrium oxygen concentration at non-standard pressure (mg L\(^{-1}\)); \( P \) is atmospheric pressure (atm); \( P_{WV} \) is partial pressure of water vapor (atm), which may be computed from:

\[
\ln P_{WV} = 11.8571 - \frac{3840.7}{T} - \frac{216961}{T^2}  \\
\varphi = 0.000975 - 1.4264 - 2.57 + 6.4364 - 8.7T^2
\]  

(3.30)

where \( T \) is temperature in °C.

### 3.5 Sediment fluxes

A schematic of the conceptual sediment model is shown in Figure 3-6. The sediment conceptual model includes fluxes and transformations of dissolved oxygen (DO), POM, NH\(_3\), NO\(_3\) and PO\(_4\). Sedimentary organic matter, which has been recently deposited, is transformed microbially and chemically into inorganic matter. This process is called diagenesis. DO, NO\(_3\), NH\(_3\) and PO\(_4\) in the sediments affect concentrations of these same constituents in the water column through mass transfer, as described in the equations above. This mass transfer is described in the equations as a diffusion process, but in reality includes other processes such as bioperturbation and tidal pumping. Coefficients were calibrated based on empirical experiments and field data, thus they do indirectly include bioperturbation and tidal pumping.
The following equations describe the sediment processes:

\[
H_p \times \frac{d\text{DO}_2}{dt} = K_{\text{DO}} \times (D\text{O}_2 - D\text{O}_2^0) - k \times \theta^{30} \times \gamma_{\text{OC}} \times P\text{OM}_2 \times H_S
\]  \hspace{1cm} (3.31)

where $H_S$ is the depth of the (active) sediment layer; $K_{\text{DO}}$ is the mass transfer rate of $\text{DO}_2$ between sediment and water column; $k$ is the decay rate of POM and $\gamma_{\text{OC}}$ is a carbon content of sediment $P\text{OM}_2$.

\[
H_p \times \frac{d\text{NH}_3}{dt} = k \times \theta^{30} \times \gamma_{\text{NH}_3} \times P\text{OM}_2 \times H_S + K_{\text{NH}_3} \times (\text{NH}_3 - \text{NH}_3^0) - k \times \theta^{30} \times \text{NH}_3^0 \times H_S
\]

Figure 3-6  Conceptual sediment model for Newport Bay.
where \( \tau_{\text{NO}_3}^{\text{S}} \) is a nitrogen content of sediment and \( k_{\text{nitrification}} \) is the nitrification rate of NH$_3$ in the sediment.

\[
H_p \times \frac{d\text{NO}_3^S}{dz} = k_{\text{nitrification}} \times \theta_{\text{tug}}^{\text{tug}} \times \text{NH}_3 \times H_p + K_{\text{NO}_3} \times (\text{NO}_3 - \text{NO}_3^S) - k_{\text{denitrification}} \times \theta_{\text{tug}}^{\text{tug}} \times \text{NO}_3 \times H_p
\]

where \( k_{\text{denitrification}} \) is the denitrification rate in the sediment.

\[
H_p \times \frac{d\text{PO}_4^S}{dt} = k \times \theta_{\text{tug}}^{\text{tug}} \times \tau_{\text{POM}} \times \text{POM} \times H_p + K_{\text{PO}_4} \times (\text{PO}_4 - \text{PO}_4^S)
\]  

(3.34)

where \( \tau_{\text{POM}} \) is a phosphorus content of sediment. All other coefficients have been previously defined.
Table 3-2  List of symbols of the water quality model

<table>
<thead>
<tr>
<th>Symbols</th>
<th>Description</th>
</tr>
</thead>
<tbody>
<tr>
<td>( M )</td>
<td>concentration of macroalgae (kg m(^{-2}))</td>
</tr>
<tr>
<td>( \mu )</td>
<td>macroalgal growth rate (day(^{-1}))</td>
</tr>
<tr>
<td>( r )</td>
<td>macroalgal respiration (and excretion) rate (day(^{-1}))</td>
</tr>
<tr>
<td>( s )</td>
<td>macroalgal scouring rate (day(^{-1}))</td>
</tr>
<tr>
<td>( m )</td>
<td>macroalgal mortality (day(^{-1}))</td>
</tr>
<tr>
<td>( f(T) )</td>
<td>function of temperature on macroalgal growth rate</td>
</tr>
<tr>
<td>( f(S) )</td>
<td>function of salinity on macroalgal growth rate</td>
</tr>
<tr>
<td>( f(L) )</td>
<td>function of light on macroalgal growth rate</td>
</tr>
<tr>
<td>( f(Q_N) )</td>
<td>function of internal nitrogen on macroalgal growth rate</td>
</tr>
<tr>
<td>( f(P) )</td>
<td>function of phosphorus on macroalgal growth rate</td>
</tr>
<tr>
<td>( f(S_P) )</td>
<td>function of space limitation on macroalgal growth rate</td>
</tr>
<tr>
<td>( f(D) )</td>
<td>function of water depth on macroalgal growth rate</td>
</tr>
<tr>
<td>( T )</td>
<td>water temperature (°C)</td>
</tr>
<tr>
<td>( T_{opt} )</td>
<td>optimum water temperature (°C)</td>
</tr>
<tr>
<td>( T_{lim} )</td>
<td>lower temperature limit below which growth ceases (°C)</td>
</tr>
<tr>
<td>( T_{max} )</td>
<td>upper temperature limit below which growth ceases (°C)</td>
</tr>
<tr>
<td>( S )</td>
<td>salinity (ppt)</td>
</tr>
<tr>
<td>( S_{opt} )</td>
<td>optimum salinity (ppt)</td>
</tr>
<tr>
<td>( S_{lim} )</td>
<td>lower salinity limit below which growth ceases (ppt)</td>
</tr>
<tr>
<td>( S_{max} )</td>
<td>upper salinity limit below which growth ceases (ppt)</td>
</tr>
<tr>
<td>( I_0 )</td>
<td>light intensity at water surface (W m(^{-2}))</td>
</tr>
<tr>
<td>( I_z )</td>
<td>light intensity at depth ( z ) (W m(^{-2}))</td>
</tr>
<tr>
<td>( h_{LS} )</td>
<td>half saturation constant for light (W m(^{-2}))</td>
</tr>
<tr>
<td>( I_{opt} )</td>
<td>optimal light level (W m(^{-2}))</td>
</tr>
<tr>
<td>( \lambda )</td>
<td>light attenuation coefficient (m(^{-1}))</td>
</tr>
<tr>
<td>( \lambda_0 )</td>
<td>non-algal portion of the light extinction coefficient (m(^{-1}))</td>
</tr>
<tr>
<td>( \lambda_1 )</td>
<td>linear macroalgal self-shading coefficient (m(^{-1}) (kg m(^{-2})) (^{-1}))</td>
</tr>
<tr>
<td>( z )</td>
<td>water depth in the water column (vertical coordinate) (m)</td>
</tr>
<tr>
<td>( k_{sp} )</td>
<td>space limitation half-saturation constant for growth (kg m(^{-2}))</td>
</tr>
<tr>
<td>( D )</td>
<td>surface water depth (m)</td>
</tr>
<tr>
<td>( h_c )</td>
<td>critical water depth for growth limitation (m)</td>
</tr>
<tr>
<td>( Q_N )</td>
<td>internal nitrogen quota of macroalgae</td>
</tr>
<tr>
<td>( Q_{N_{min}} )</td>
<td>minimum internal nitrogen quota</td>
</tr>
<tr>
<td>( Q_{N_{max}} )</td>
<td>maximum internal nitrogen quotas for macroalgae</td>
</tr>
<tr>
<td>( k_p )</td>
<td>nitrogen half-saturation constant for growth limitation</td>
</tr>
<tr>
<td>( k_{P_{min}} )</td>
<td>phosphate half-saturation constant for growth limitation (mg m(^{-3}))</td>
</tr>
<tr>
<td>( U )</td>
<td>current speed (cm s(^{-1}))</td>
</tr>
<tr>
<td>( U_{min} )</td>
<td>minimum current speed at which sloughing occurs (cm s(^{-1}))</td>
</tr>
<tr>
<td>( U_{max} )</td>
<td>maximum current speed at which sloughing of all the macroalgae occurs (cm s(^{-1}))</td>
</tr>
</tbody>
</table>
maximum mortality rate at the optimum temperature (day\(^{-1}\))

### Chemical symbols

<table>
<thead>
<tr>
<th>Symbol</th>
<th>Description</th>
</tr>
</thead>
<tbody>
<tr>
<td>(\text{NO}_3)</td>
<td>concentration of nitrate (mg m(^{-3}))</td>
</tr>
<tr>
<td>(k_{nitr})</td>
<td>nitrification rate of (\text{NH}_3) (day(^{-1}))</td>
</tr>
<tr>
<td>(k_{den})</td>
<td>denitrification rate of (\text{NO}_3) (day(^{-1}))</td>
</tr>
<tr>
<td>(\Theta_{TM})</td>
<td>non-dimensional temperature multiplier of reaction</td>
</tr>
<tr>
<td>(k_{trans})</td>
<td>mass transfer rate of (\text{NO}_3) (m day(^{-1}))</td>
</tr>
<tr>
<td>(V_{NO})</td>
<td>nitrate uptake rate (mg-N dwg(^{-1}) d(^{-1}))</td>
</tr>
<tr>
<td>(V_{NO,max})</td>
<td>maximum nitrate uptake rate (mg-N dwg(^{-1}) d(^{-1}))</td>
</tr>
<tr>
<td>(k_{NO})</td>
<td>halfsaturation constant for (\text{NO}_3) (mg m(^{-3}))</td>
</tr>
<tr>
<td>(\eta_{nitr})</td>
<td>The nitrification inhibition factor</td>
</tr>
<tr>
<td>(k_{fri2})</td>
<td>first-order nitrification inhibition coefficient</td>
</tr>
<tr>
<td>(\text{NH}_3)</td>
<td>concentration of ammonia (mg m(^{-3}))</td>
</tr>
<tr>
<td>(V_{NH}_3)</td>
<td>ammonia uptake rate (mg-N dwg(^{-1}) d(^{-1}))</td>
</tr>
<tr>
<td>(k_{transNH})</td>
<td>mass transfer rate of (\text{NH}_3) (mg-N dwg(^{-1}) d(^{-1}))</td>
</tr>
<tr>
<td>(V_{NH}_3,max)</td>
<td>maximum ammonia uptake rate (mg-N dwg(^{-1}) d(^{-1}))</td>
</tr>
<tr>
<td>(k_{NH}_3)</td>
<td>half saturation constant for (\text{NH}_3) (mg m(^{-3}))</td>
</tr>
<tr>
<td>(I_{N})</td>
<td>internal nitrogen (kg m(^{-2}))</td>
</tr>
<tr>
<td>(\text{PO}_4)</td>
<td>orthophosphate concentration (mg m(^{-3}))</td>
</tr>
<tr>
<td>(e_\text{p})</td>
<td>fraction of macroalgae that is phosphorus</td>
</tr>
<tr>
<td>(k_{transPO})</td>
<td>mass transfer rate of (\text{PO}_4) (m day(^{-1}))</td>
</tr>
<tr>
<td>(k_{ads})</td>
<td>rate of adsorption onto suspended solids (day(^{-1}))</td>
</tr>
<tr>
<td>(\text{POM})</td>
<td>particulate organic matter (mg m(^{-3}))</td>
</tr>
<tr>
<td>(\text{DO})</td>
<td>dissolved oxygen concentration (mg L(^{-1}))</td>
</tr>
<tr>
<td>(\Theta_{DO})</td>
<td>non-dimensional temperature multiplier of DO</td>
</tr>
<tr>
<td>(\alpha_\text{p})</td>
<td>reaeration rate (day(^{-1}))</td>
</tr>
<tr>
<td>(\alpha_{2})</td>
<td>saturation concentration of oxygen at non-standard pressure (mg L(^{-1}))</td>
</tr>
<tr>
<td>(r_{OC})</td>
<td>ratio of mass of oxygen consumed per mass of carbon assimilated</td>
</tr>
<tr>
<td>(e_c)</td>
<td>fraction of macroalgae that is carbon</td>
</tr>
<tr>
<td>(k_{transDO})</td>
<td>mass transfer rate of (\text{DO}) between sediment and water column (m day(^{-1}))</td>
</tr>
<tr>
<td>(r_{ON})</td>
<td>ratio of mass of oxygen consumed per mass of nitrogen oxidized</td>
</tr>
<tr>
<td>(U_{m})</td>
<td>mass-transfer velocity in the water layer (m d(^{-1}))</td>
</tr>
<tr>
<td>(U_{tv})</td>
<td>mean tidal velocity over a complete tidal cycle (m day(^{-1}))</td>
</tr>
<tr>
<td>(\text{O}_{eq})</td>
<td>equilibrium oxygen concentration at 1 atm (mg L(^{-1})) within the range 0.0 to 40.0 °C</td>
</tr>
<tr>
<td>(T_e)</td>
<td>water temperature (°K)</td>
</tr>
<tr>
<td>(P)</td>
<td>atmospheric pressure (atm)</td>
</tr>
<tr>
<td>(P_{uv})</td>
<td>partial pressure of water vapor (atm)</td>
</tr>
</tbody>
</table>

### Sediment symbols

<table>
<thead>
<tr>
<th>Symbol</th>
<th>Description</th>
</tr>
</thead>
<tbody>
<tr>
<td>(\text{POM}_s)</td>
<td>settled POM in the sediments (mg m(^{-2}))</td>
</tr>
<tr>
<td>(h_{s})</td>
<td>depth of biologically-active sediment layer (m)</td>
</tr>
<tr>
<td>(k_{transDO})</td>
<td>mass transfer rate of (\text{DO}) between sediment and water column (m day(^{-1}))</td>
</tr>
<tr>
<td>(k)</td>
<td>decay rate of (\text{POM}) (day(^{-1}))</td>
</tr>
<tr>
<td>Symbol</td>
<td>Description</td>
</tr>
<tr>
<td>--------</td>
<td>--------------------------------------------------</td>
</tr>
<tr>
<td>$\tau_{Sc}$</td>
<td>carbon content of sediment $POM_{2}$</td>
</tr>
<tr>
<td>$\tau_{SN}$</td>
<td>nitrogen content of sediment $POM_{2}$</td>
</tr>
<tr>
<td>$k_{nitr}$</td>
<td>nitrification rate of $NH_{3}$ in the sediment (day^{-1})</td>
</tr>
<tr>
<td>$k_{denit}$</td>
<td>denitrification rate in the sediment (day^{-1})</td>
</tr>
<tr>
<td>$\tau_{PF}$</td>
<td>phosphorus content of sediment $PPM_{5}$</td>
</tr>
<tr>
<td>$DO_{s}$</td>
<td>concentration of dissolved oxygen in the sediment (mg m^{-3})</td>
</tr>
<tr>
<td>$NH_{3}$</td>
<td>concentration of ammonia in the sediment (mg m^{-3})</td>
</tr>
<tr>
<td>$NO_{3}$</td>
<td>concentration of nitrate in the sediment (mg m^{-3})</td>
</tr>
<tr>
<td>$PO_{4}$</td>
<td>concentration of phosphate in the sediment (mg m^{-3})</td>
</tr>
<tr>
<td>Symbols</td>
<td>Description</td>
</tr>
<tr>
<td>---------</td>
<td>-------------</td>
</tr>
<tr>
<td>$\mu_{\text{max}}$</td>
<td>maximum growth rate at the optimum temperature</td>
</tr>
<tr>
<td>$T_{\text{opt}}$</td>
<td>optimum temperature at which growth rate is maximum</td>
</tr>
<tr>
<td>$T_{\text{min}}$</td>
<td>lower temperature limit below which growth ceases</td>
</tr>
<tr>
<td>$T_{\text{max}}$</td>
<td>upper temperature limit above which growth ceases</td>
</tr>
<tr>
<td>$S_{\text{opt}}$</td>
<td>optimum salinity at which growth rate is maximum</td>
</tr>
<tr>
<td>$S_{\text{min}}$</td>
<td>lower salinity limit at which growth rate ceases</td>
</tr>
<tr>
<td>$S_{\text{max}}$</td>
<td>upper salinity limit at which growth ceases</td>
</tr>
<tr>
<td>$I_{\text{opt}}$</td>
<td>optimum (saturating) light intensity</td>
</tr>
<tr>
<td>$\alpha_{\text{al}}$</td>
<td>non-algal portion of the light extinction coefficient</td>
</tr>
<tr>
<td>$\alpha_{\text{la}}$</td>
<td>linear macroalgae self-shading coefficient</td>
</tr>
<tr>
<td>$k_{\text{sp}}$</td>
<td>space limitation half-saturation constant for growth limitation</td>
</tr>
<tr>
<td>$k_{\text{cr}}$</td>
<td>critical water depth for growth limitation</td>
</tr>
<tr>
<td>$k_{\text{hS}}$</td>
<td>half saturation constant for $N_{\text{IN}}$</td>
</tr>
<tr>
<td>$q_{\text{IN}}$</td>
<td>minimum internal nitrogen quota</td>
</tr>
<tr>
<td>$q_{\text{INmax}}$</td>
<td>maximum internal nitrogen quota</td>
</tr>
<tr>
<td>$V_{\text{N}}$</td>
<td>maximum nitrate uptake rate</td>
</tr>
<tr>
<td>$k_{\text{hS}}$</td>
<td>half saturation constant for $N_{\text{NH3}}$</td>
</tr>
<tr>
<td>$V_{\text{NH3}}$</td>
<td>maximum ammonia uptake rate</td>
</tr>
<tr>
<td>$k_{\text{hS}}$</td>
<td>half saturation constant for $N_{\text{PO4}}$</td>
</tr>
<tr>
<td>$k_{\text{hS}}$</td>
<td>phosphate half-saturation constant for growth limitation</td>
</tr>
<tr>
<td>$V_{\text{max}}$</td>
<td>maximum respiration rate at the optimum temperature</td>
</tr>
<tr>
<td>$V_{\text{min}}$</td>
<td>minimum current speed for sloughing</td>
</tr>
<tr>
<td>Parameter</td>
<td>Value</td>
</tr>
<tr>
<td>-----------</td>
<td>-------</td>
</tr>
<tr>
<td>Maximum current speed for total sloughing</td>
<td>82-128</td>
</tr>
<tr>
<td>Maximum mortality rate at the optimum temperature</td>
<td>0.003-0.8</td>
</tr>
<tr>
<td>Fraction of macroalgae that is phosphorus</td>
<td>0.0004-0.0086</td>
</tr>
<tr>
<td>Fraction of macroalgae that is carbon</td>
<td>0.27-0.36</td>
</tr>
</tbody>
</table>

**Chemical processes**

<table>
<thead>
<tr>
<th>Parameter</th>
<th>Value</th>
<th>Units</th>
<th>Reference</th>
</tr>
</thead>
<tbody>
<tr>
<td>Nitrification rate of</td>
<td>0.001-1.3</td>
<td>day^{-1}</td>
<td>Bowie et al. (1985)</td>
</tr>
<tr>
<td>Denitrification rate of</td>
<td>0.0-1.0</td>
<td>day^{-1}</td>
<td>Bowie et al. (1985)</td>
</tr>
<tr>
<td>Temperature coefficient for oxygen transfer</td>
<td>1.045-1.08</td>
<td>day^{-1}</td>
<td>Bowie et al. (1985)</td>
</tr>
<tr>
<td>First-order nitrification inhibition coefficient</td>
<td>0.6</td>
<td>-</td>
<td>Chapra (1997)</td>
</tr>
<tr>
<td>Rate of adsorption of phosphate onto suspended solids</td>
<td>-</td>
<td>day^{-1}</td>
<td>Chapra (1997)</td>
</tr>
<tr>
<td>Temperature coefficient for SOD</td>
<td>1.04-1.13</td>
<td>day^{-1}</td>
<td>Chapra (1997)</td>
</tr>
</tbody>
</table>

**Sediment processes**

<table>
<thead>
<tr>
<th>Parameter</th>
<th>Value</th>
<th>Units</th>
<th>Reference</th>
</tr>
</thead>
<tbody>
<tr>
<td>Decay rate of POM</td>
<td>0.0018-0.035</td>
<td>day^{-1}</td>
<td>DiToro (2001)</td>
</tr>
<tr>
<td>Depth of biologically-active sediment layer</td>
<td>-</td>
<td>m</td>
<td></td>
</tr>
<tr>
<td>Nitrification rate of ammonia in the sediment</td>
<td>-</td>
<td>day^{-1}</td>
<td></td>
</tr>
<tr>
<td>Denitrification rate in the sediment</td>
<td>-</td>
<td>-</td>
<td></td>
</tr>
<tr>
<td>Mass transfer rate of DO in the sediment</td>
<td>-</td>
<td>m day^{-1}</td>
<td>Sanford and Crawford(2000)</td>
</tr>
<tr>
<td>Mass transfer rate of N in the sediment</td>
<td>0.06-0.1464</td>
<td>m day^{-1}</td>
<td>Sanford and Crawford(2000)</td>
</tr>
<tr>
<td>Mass transfer rate of P in the sediment</td>
<td>0.0816</td>
<td>m day^{-1}</td>
<td>Sanford and Crawford(2000)</td>
</tr>
<tr>
<td>Carbon content of sediment POM</td>
<td>0.0096-0.012</td>
<td>-</td>
<td>Peng et al.(2007); Sutula et al.(2006)</td>
</tr>
<tr>
<td>Nitrogen content of sediment POM</td>
<td>0.0008-0.0012</td>
<td>-</td>
<td>Peng et al.(2007); Sutula et al.(2006)</td>
</tr>
<tr>
<td>Phosphorus content of sediment POM</td>
<td>0.00048-0.00054</td>
<td>-</td>
<td>Sutula et al.(2006)</td>
</tr>
</tbody>
</table>
Since only one literature value is available, a calibrated value within reasonable range of the observed value was considered acceptable.

Literature values are not available.

Considering spatial variation of the bay, two values were used in the simulation.

Model results were improved by using a value outside the range of literature values.
Table 3-4. Summary of differential equations for the constituents of the water quality model

Macroalgae

\[
\frac{dM}{dt} = (\mu - r - s - m) \times M \quad (3.1)
\]

\[
\mu = \mu_{\text{max}} \times f(T) \times f(S) \times f(I) \times f(SP) \times f(D) \times \text{min} \left( f(QN), f(P) \right) \quad (3.2)
\]

\[
f(T) = \exp \left[ -2.3 \times \left( \frac{T - T_{\text{opt}}}{T_{\text{opt}} - T_{\text{min}}} \right)^2 \right] \quad (3.3)
\]

\[\begin{align*}
&\text{if } T < T_{\text{opt}} \quad T_{\nu} = T_{\text{min}} \\
&\text{if } T > T_{\text{opt}} \quad T_{\nu} = T_{\text{max}}
\end{align*}\]

\[
f(S) = \begin{cases} 
1 - \left( \frac{S - S_{\text{opt}}}{S_{\text{opt}} - S_{\text{min}}} \right)^m & \text{for } S \geq 5 \\
\left( \frac{S - S_{\text{opt}}}{S_{\text{opt}} - S_{\text{min}}} \right)^m & \text{for } S < 5
\end{cases} \quad (3.4)
\]

\[\text{if } S < S_{\text{opt}} \quad S_{\nu} = S_{\text{min}} \text{ and } m = 2.5\]

\[
f(I) = \frac{k_1}{k_0} \exp \left( \frac{-I}{3} \right) \quad (3.5)
\]

\[I_0 = I_0 \exp (-\lambda I_0) \quad (3.8)
\]

\[
\lambda = \lambda_0 + \lambda_1 M \quad (3.9)
\]

\[
f(SP) = 1 - \frac{M}{k_{sp}} \quad (3.10)
\]

\[f(D) = \begin{cases} 
1 & \text{if } D > k_D f(D) = 0 & \text{if } D < k_D
\end{cases} \quad (3.11)
\]

\[
f(QN) = \frac{Q(N - 2N_{\text{min}})}{k_{QN}QW + Q_{\text{min}}} \quad (3.12)
\]

\[
f(P) = \frac{\mu \sigma}{k_{PQ} + 4} \quad (3.13)
\]

\[r = r_{\text{max}} \times f(T) \quad (3.14)\]
\[ s = \frac{U - U_{\text{min}}}{U_{\text{max}} - U_{\text{min}}} \quad \text{if} \; U > U_{\text{min}} \]  
\[ m = m_{\text{max}} \times \left[ 1 - \left( \frac{f(\text{J}) \times f(\text{L}) \times \min \left( f(Q_N), f(P) \right)}{f(Q_N), f(P)} \right) \right] \]  

Nitrogen

\[ \frac{dn_\text{NO}_3}{dt} = (k_{\text{nit}} \times NH_3 \times f_{\text{nit}} - k_{\text{dmt}} \times NO_3) \times \theta_{\text{pH}} - V_{\text{NO}} \times \frac{M}{D} - K_{\text{nNNO}}(NO_3 - NO_3_2) \]  
\[ V_{\text{NO}} = V_{\text{mNO}} \times \frac{NO_3}{k_{\text{NH}_2} + NO_3} \times \frac{Q_N \max - Q_N}{Q_N \max - Q_N \min} \]  
\[ f_{\text{nit}} = 1 - \exp \left( -k_{\text{nNNO}} \times DO \right) \]  
\[ \frac{dn\text{NH}_3}{dt} = r \times Q_N \times \frac{M}{D} - V_{NH} \times \frac{M}{D} - k_{\text{nht}} \times NH_3 \times f_{\text{nht}} \times \theta_{\text{pH}} - K_{\text{SNH}}(NH_3 - NH_3_2) \]  
\[ V_{NH} = V_{\text{mNH}} \times \frac{NH_3}{k_{\text{NH3}} + NH_3} \times \frac{Q_N \max - Q_N}{Q_N \max - Q_N \min} \]  
\[ \frac{dn_\text{IN}}{dt} = (V_{\text{NO}} + V_{NH}) \times M - (r + s + m) \times IN \]  

Phosphorus

\[ \frac{dp_\text{PO}_4}{dt} = (r - \mu) \times \alpha_p \times \frac{M}{D} - K_{\text{dPO}}(PO_4 - PO_4_2) - k_{\text{nads}} \times PO_4 \times \theta_{\text{pH}} \]  

DO

\[ \frac{dD}{dt} = k_a \times \theta_{\text{DO}} \times \left( O_2 - DO \right) + (\mu - r) \times n_{\text{DO}} \times \alpha_c \times \frac{M}{D} \]  
\[ -K_{\text{dDO}}(DO - DO_2) - k_{\text{nht}} \times n_{\text{NH3}} \times NH_3 \times f_{\text{nht}} \times \theta_{\text{pH}} \]  
\[ k_a = 3.93 \frac{\sqrt{c}}{D^{1/3}} + 0.7390 \times 10^{-5} \frac{1}{D} + 6.087 \]  
\[ \ln Q_{\text{scr}} = -139.34410 + \frac{1.6787017}{r_{\text{th}}} - \frac{6.5428038 \times 10^{-7}}{r_{\text{th}}^2} + \frac{1.2483090e10}{r_{\text{th}}^4} - \frac{8.62149e12}{r_{\text{th}}^6} - \frac{1}{1.90625} \times S \times \left[(3.1929e - 2) - \left( \frac{1.9428e + 1}{r_{\text{th}}} \right) \left( \frac{1.6672e + 1}{r_{\text{th}}} \right) \right] \]  
\[ \text{OS} = Q_{\text{scr}} \frac{(1 - P + P \phi)(1 - \phi P)}{(1 - P \phi P)(1 - \phi P)} \]
\begin{equation}
\ln P_{W} = 11.8671 - \frac{384.03}{T} - \frac{713.441}{T^2} ;
\end{equation}
\begin{equation}
\varphi = 0.000975 - 1.426e - 5T + 6.436e - 6T^2
\end{equation}

\textbf{Sediment}

\begin{equation}
H_g \times \frac{dDO_2}{dt} = K_{SDO} \times (DO - DO_2) - k \times \theta_{POM}^{T-\alpha} \times r_{SC} \times POM_g \times H_g
\end{equation}

\begin{equation}
H_g \times \frac{dNH_3}{dt} = k \times \theta_{POM}^{T-\alpha} \times r_{SN} \times POM_g \times H_g + K_{SNH3} \times (NH_3 - NH_3_g)
\end{equation}
\begin{equation}
- k_{PN} \times \theta_{POM}^{T-\alpha} \times NH_3_2 \times H_g
\end{equation}

\begin{equation}
H_g \times \frac{dNO_3}{dt} = k_{SN} \times \theta_{POM}^{T-\alpha} \times NH_3_2 \times H_g + K_{SNNO3} \times (NO_3 - NO_3_g)
\end{equation}
\begin{equation}
- k_{PN} \times \theta_{POM}^{T-\alpha} \times NO_3 \times H_g
\end{equation}

\begin{equation}
H_g \times \frac{dPO_4}{dt} = k \times \theta_{POM}^{T-\alpha} \times r_{SP} \times POM_g \times H_g + K_{SPC4} \times (PO_4 - PO_4_g)
\end{equation}
4. Description of hydrodynamic model (RMA2) and sediment transport and water quality model (RMA11)

The RMA suite of finite element hydrodynamic, sediment transport and water quality models employed for this study have been used extensively since 1977 in engineering applications to examine flow and transport of constituents in surface water systems (RMA, 1997; 1998a; 1998b; 1999; 2001; 2003). One of the unique characteristics of the suite of models is the ability to represent a physical system using 1, 2, and/or 3 dimensional approximations within a single computational network. This allows construction of efficient computational networks where the level of spatial resolution varies according to the needs of the problem. Originally developed with the support of the U.S. Army Corps of Engineers Waterways Experiments Station, the models have undergone continued development and refinement by RMA. One of the most important additions has been the capability to accurately represent wetting and drying in shallow estuaries (RMA, 2001).

Hydrodynamics have been simulated for this study using RMA-2, a two-dimensional depth-averaged finite element model that solves the shallow water equations to provide temporal and spatial descriptions of velocities and water depths throughout the regions of interest. The model uses the Smagorinski formulation for modeling of turbulent momentum transfer (King et al., 1975). RMA-2, capable of simulating the de-watering of tidal flats, is uniquely suited for modeling of inter-tidal hydrodynamics in the marshes and mudflats that characterize boundaries of Upper Newport Bay.

Sediment transport and water quality are simulated using the finite element water quality model, RMA-11, which solves the mass transport equation in divergence or non-divergence form for multiple non-linearly coupled constituents. RMA-11 has been successfully applied in numerous previous projects to simulate the fate and transport of sediments and other conservative and non-conservative water quality constituents in surface water systems. Velocities and water depths obtained from hydrodynamic model results are used to solve the advection-dispersion equation for each constituent simulated. RMA-11 has been designed for compatibility with model results obtained from one-, two-, or three-dimensional hydrodynamic simulations (US Army Corps of Engineers, 1997; 1998).

4.1 Model Configuration

The RMA finite element model of Newport Bay, shown in Figure 4-1, extends from the tidal boundary at the entrance to the Lower Bay to San Diego Creek just upstream from Jamboree Bridge. Tributary inflows include San Diego Creek and Santa Ana Delhi Channel. The Bay is represented using a two-dimensional depth averaged approximation, with short one-dimensional
cross-sectionally averaged segments at the tidal boundary, San Diego Creek, and Santa Ana Delhi Channel. Even though 2-D model cannot simulate vertical variations in dissolved oxygen or other water quality parameters, modeling studies have focused on the shallow Upper Bay where stratification is not typically important, especially during storm seasons, and depth-average representation of flow and water quality constituent concentrations provides a good approximation for most of the year.

4.1.1 Bathymetry

The finite element mesh includes the area of the Bay up to approximately +1 m (3.3 ft) MSL. Because the bed elevations in the Bay change over time, several bathymetric data sets have been developed for the model. The grid used for this study was developed using the most recent bathymetric data, which is from the comprehensive bathymetry survey performed by the County of Orange and the Unit III dredging design plans (County of Orange Public Facilities and Resources Department, 1997; unpublished data from U.S. Army Corps of Engineering, 2004; 2006). The individual data sets do not cover the entire Bay, therefore, at any given location in the finite element mesh, the most recent available data were used to set the elevations. The 2006 data (collected prior to the most recent dredging project) were used to set elevations for all areas above PCH Bridge except for the Newport Dunes. The most recent data for the Newport Dunes was collected in 2004. The Lower Bay elevations were set using the 1997 data. A color contour plot of the model bathymetry is shown in Figure 4-2.

4.1.2 Boundary conditions

Boundary conditions for hydrodynamics that are required for dynamic simulations include tidal elevations at the ocean boundary and tributary inflows to the system. A steady or time-varying surface wind, with a specified direction, may be applied to examine the effects of wind stress on circulation in the system, however wind has not been applied in this study.

<table>
<thead>
<tr>
<th>Boundary</th>
<th>Constituents for boundary conditions</th>
</tr>
</thead>
<tbody>
<tr>
<td>Ocean</td>
<td>Tide elevation, temperature, salinity, DO, NO3, NH3, PO4</td>
</tr>
<tr>
<td>Tributaries</td>
<td>Flow rate, temperature, salinity, DO, NO3, NH3, PO4</td>
</tr>
<tr>
<td>Sediment</td>
<td>Deposited sediments (from sediment transport model result)</td>
</tr>
<tr>
<td>Atmosphere</td>
<td>Air temperature, air pressure, vapor pressure, solar radiation, cloudiness</td>
</tr>
</tbody>
</table>
Water quality simulations require specification of constant or time-varying concentrations at the San Diego Creek and Santa Ana Delhi inflow boundaries, and the tidal boundary for all water column constituents simulated. Meteorological data inputs are also required, including solar radiation, air temperature, wind speed and vapor pressure. These data can be applied over the entire model or varied by element type.

Table 4-1 shows hydrodynamic and water quality constituents used for each boundary condition for this study.

Sediment transport simulations require specification of sediment concentrations for the San Diego Creek and Santa Ana Delhi inflows. Tidal boundary sediment concentrations are set to zero.

Results from the sediment transport model serve as an additional boundary condition for the water quality model. The sediment depth and distribution are fed into the water quality model to calculate nutrients available for flux into the water column.

4.2 Previous Model Calibration

The model of Newport Bay was previously calibrated for hydrodynamics, salinity transport and sediment transport during the initial studies for U.S. Army Corp of Engineers (RMA, 1997, 1998a and 1998b). No additional calibration of hydrodynamics or transport is performed as part of this study as recent velocity data were not available and the minor bathymetry modifications would not affect stage or transport.

The hydrodynamic model was calibrated to observed stage time-series and spot velocity measurements from the 1992 – 1993 Reconnaissance Study (Coastal Frontiers Corporation, 1993).

The salinity model was calibrated separately for wet and dry weather periods using salinity observations in Newport Bay including time-series of near-surface salinity collected by IRWD and spot measurements of salinity profiles over the water column collected by IRWD and the County of Orange during November 1996 and August – September 1997.
Figure 4-1  Finite element model configuration of Newport Bay, with monthly water quality monitoring stations of the Orange County where UNBJAM is Upper Newport Bay – Unit 1 Basin; UNBSDC is Upper Newport Bay – Unit 2 Basin; UNBNSB is Upper Newport Bay – North Star Beach; UNBCHB is Upper Newport Bay – Coast Highway Bridge.
Figure 4-2  Color contours of the Newport Bay model bathymetry.
5. Water quality model calibration and validation

To calibrate the water quality model, variations in water quality throughout the Bay were computed using time varying water quality boundary conditions and daily averaged tributary inflows for San Diego Creek and Santa Ana Delhi Channel, the tide corresponding to the calibration period, and meteorological data from the California Irrigation Management System (CIMIS) station at Irvine and from the weather station at John Wayne Airport.

The objective of the water quality model calibration is to reproduce the depth average of observed temperature, dissolved oxygen, macroalgae and nutrient values in the Upper Newport Bay, and the seasonal macroalgal growth patterns. Calibration of the water quality model involves adjustment of model parameters to best represent temperature, DO, macroalgae and nutrient concentrations throughout the system. The following constituents are included in the water quality model.

- Salinity
- Temperature
- DO
- Ammonia (NH3)
- Nitrate (NO3)
- Phosphate (PO4)
- Macroalgae
- Macroalgal internal nitrogen
- Deposited particulate organic matter (POM)

The constituents of the water quality model exclude dissolved organic nitrogen (DON) since the time series of DON is not available. Therefore, total nitrogen (TN) is assumed to consist mainly of dissolved inorganic nitrogen and particulate organic nitrogen in this study.

The primary goal of model validation is to assure that the simulated macroalgal growth patterns follow the observed variations from year to year that are presumed to result from variations in wet season inflows (and associated suspended sediment) to the Bay.

Calibration of the model was a complicated process involving adjustment of many interrelated parameters. Adjustment of one parameter could improve the macroalgae calibration result, but at the same worsen results for one of the nutrients in the water column. Improvement in DO calibration could result in worse macroalgae calibration, and so on. As such, model parameters were adjusted until a balance was achieved with good calibration results for all constituents.
Although there are many unknowns in the system, this balance gives good indication that all model parameters are reasonably well calibrated.

5.1 Calibration and validation periods
The model calibration period is from January through December 2004. The validation period is from January 2005 through December 2006. These periods were chosen because they include a year with a dryer winter and smaller blooms later in the year (2003-2004), and a year with large winter flows and subsequent larger macroalgal blooms later in the year (2004 – 2005). The validation is extended into 2006 to cover another dryer year. Sediment inflow data are only available through June 2006, however the full year was still simulated as for any given year, there is typically very little sediment deposited after June 1st until wet weather arrives again at the end of the year.

If model parameters calibrated for a dry year can also produce a good match with observed data during a wet year, this provides much better assurance of model validation than calibrating and validating for similar year types.

5.2 Boundary conditions
Model boundary conditions are set for the January 2004 through December 2006 calibration and validation period using the following data sets.

- San Diego Creek and Santa Ana Delhi daily inflows and two- to four-times monthly electrical conductivity (EC), NH3, NO3, PO4, DO, and total suspended solids (TSS) concentrations: Nutrient TMDL reports (County of Orange et al., 2004; 2005; 2006, [http://www.ocwatersheds.com/watersheds/tmdls.asp](http://www.ocwatersheds.com/watersheds/tmdls.asp)).
- Downstream tidal boundary: National Oceanic and Atmospheric Administration (NOAA) 6-minute predicted tide at Newport Harbor entrance ([http://tidesandcurrents.noaa.gov](http://tidesandcurrents.noaa.gov)).
- Tidal boundary temperature and salinity from Newport Pier in Newport Beach: shore station program at the University of California at San Diego ([http://shorestation.ucsd.edu/active/index_active.html](http://shorestation.ucsd.edu/active/index_active.html)).
- Tidal boundary DO, NO3, PO4 and phytoplankton (chlorophyll-a): California Cooperative Oceanic Fisheries Investigations ([http://www.calcofi.org](http://www.calcofi.org)). Boundary conditions for remaining constituents, for which no data were available, were estimated.
- Hourly solar radiation and vapor pressure: California Irrigation Management System (CIMIS) station at Irvine ([http://www.cimis.water.ca.gov](http://www.cimis.water.ca.gov)).
• Hourly air temperature and wind speed at John Wayne Airport: MesoWest (http://www.met.utah.edu/mesowest/).

TKN data from San Diego Creek and Santa Ana Delhi was not used for model boundary conditions because it provides no information on particulate vs. dissolved fractions. Instead the particulate fraction was estimated using TSS and assumptions about TOC and % nitrogen of TOC. The dissolved organic nitrogen component of TKN was assumed to be not significant since the TN concentrations from inflows matched the sum of observed data of NO3, NH4 and particulate organic nitrogen (% nitrogen of TSS).

Daily creek inflows are plotted in Figure 5-2 for the period of December 2003 through the end of 2006. To more clearly display the summertime flows, a log scale plot is also provided in Figure 5-3.

Many of the water quality data sets are limited and estimation of boundary conditions was required. Maps of the data collections stations are shown in Figure 5-1 (maps from TMDL report, Orange County et. al, 2004).

The daily concentrations of water quality constituents were estimated by non-linear regressions between available data for the constituents and daily flow rates multiplied by daily air temperatures. Regressions using daily flows and air temperatures are useful because these data sets are complete. Inclusion of air temperature in the regression equations brings seasonality into the estimated time series. Daily water temperatures were estimated by the following regression equation:

\[
\log_{10}(\text{water temperature} \times \text{flow rate}) = a + b \times \log_{10}(\text{air temperature} \times \text{flow rate})
\]  

(4.1)

where \(a\) and \(b\) are defined in Table 5-1.

Daily DO values were estimated using the following regression equation:

\[
\log_{10}(\text{DO} \times \text{air temperature}) = a + b \times \log_{10}(\text{air temperature} \times \text{flow rate}) + c \times \log_{10}(\text{air temperature} \times \text{flow rate})^a
\]

where \(a\) and \(b\) are defined in Table 5-1.
As with daily DO, daily nutrients were estimated using non-linear base-10 logarithmic regressions between nutrients $X$ daily temperature and daily flow rates $X$ air temperatures.

$$\log_{10}(nutrients \times \text{air temperature}) = a + b \times \log_{10}(\text{air temperature} \times \text{flow rate}) + c \times \log_{10}(\text{air temperature} \times \text{flow rate})^2$$

(4.3)

where $a$ and $b$ are defined in

Table 5-1. For consistency of generating input files, the same strategy was used for estimating all nutrient model input time series.

Final boundary conditions were input to the model as daily averages and included the observed data points with estimated values filling in missing data. Plots of observed, estimated and model boundary condition water temperature, DO, NO3, NH3 and PO4 for San Diego Creek and Santa Ana Delhi Channel are shown in Figure 5-4 through Figure 5-13 for the 2004 through 2006 calibration/validation period. Note that observed data values are affected by detection limits and significant figures in reported data, particularly for NH3 and PO4.

| Table 5-1  Constants of non-linear regression for water quality constituents of inflows |
|-----------------|--------|--------|--------|--------|--------|--------|
|                | a      | b      | c      | a      | b      | c       |
| Water temperature | SAD    | SDC    | SAD    | SDC    | SAD    | SDC     |
| 1.007626       | 0.993695 | 0.07879 | 0.091694 |         |        |         |
| EC              | 0.137876 | -0.069 | 0.211749 | 1.09074 | 3.020371 | 2.319728 |
| DO              | -0.02843 | -0.09291 | 0.002322 | 0.4763 | 2.362099 | 1.646   |
EC was not modeled, but used to compute boundary conditions for salinity. EC was estimated using the following regression equation:

\[
\log_{10}(EC \times \text{flow rate}) = a + b \times \log_{10}(\text{air temperature} \times \text{flow rate}) + c \\
\times \log_{10}(\text{air temperature} \times \text{flow rate})^d
\]

where \(a\) and \(b\) are defined in Table 5-1.

Salinity was calculated from the combined observed and estimated EC, shown in Figure 5-14 and Figure 5-15, using a regression between EC and salinity as follows:

\[
\text{Salinity} = a + b \times EC
\]

(4.5)

where \(a\) and \(b\) are constants set as values of 2.18 and 0.554062, respectively. The regression was obtained from one of the Sacramento – San Joaquin delta channels with a similar range of salinity (e.g., one of which EC minimum is 129 and EC maximum is 7100 µmhos/cm). Estimated salinity of the two inflows is presented in Figure 5-16 and Figure 5-17. The regressions of water quality constituents of Santa Ana Delhi and San Diego Creek are shown in from Figure 5-18 to Figure 5-27.
Daily temperature and salinity for the ocean boundary of Newport Bay were obtained from the shore station program at the University of California at San Diego (UCSD) (http://shorestation.ucsd.edu/active/index_active.html).

Water quality data for the ocean boundary are available at LA harbor with a sampling interval of four times a year from the California Cooperative Oceanic Fisheries Investigation (CalCOFI) (http://www.calcofi.org/). Available data include NO3, PO4, DO and chlorophyll - a. Averages of data for 2004 through 2006 were used. Ocean boundary conditions for the remaining constituents (NH3 and VSS) were set as constants based on Southern California Coastal Water Research Project (SCCWRP) measurements (via personal communication).

Table 5-2 summarizes the data sources and missing data estimation methods for all model boundary conditions.
Table 5-2  Summary of data sources and estimation methods for water quality model constituent boundary conditions.

<table>
<thead>
<tr>
<th>Constituent</th>
<th>Data Source</th>
<th>Estimation Method for Missing Data</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>San Diego Creek</td>
<td>Santa Ana Delhi</td>
</tr>
<tr>
<td>Flow/stage</td>
<td>Nutrient TMDL</td>
<td>Nutrient TMDL</td>
</tr>
<tr>
<td></td>
<td></td>
<td>No missing data</td>
</tr>
<tr>
<td>Temperature</td>
<td>Nutrient TMDL</td>
<td>Nutrient TMDL</td>
</tr>
<tr>
<td></td>
<td></td>
<td>Air/water temperature, flow non-linear regression</td>
</tr>
<tr>
<td>Salinity</td>
<td>Nutrient TMDL</td>
<td>Nutrient TMDL</td>
</tr>
<tr>
<td></td>
<td></td>
<td>EC linear regression</td>
</tr>
<tr>
<td>DO</td>
<td>Nutrient TMDL</td>
<td>Nutrient TMDL</td>
</tr>
<tr>
<td></td>
<td></td>
<td>Air temperature, flow non-linear regression</td>
</tr>
<tr>
<td>NH3</td>
<td>Nutrient TMDL</td>
<td>Nutrient TMDL</td>
</tr>
<tr>
<td></td>
<td></td>
<td>Air temperature, flow non-linear regression</td>
</tr>
<tr>
<td>NO3</td>
<td>Nutrient TMDL</td>
<td>Nutrient TMDL</td>
</tr>
<tr>
<td></td>
<td></td>
<td>Air temperature, flow non-linear regression</td>
</tr>
<tr>
<td>PO4</td>
<td>Nutrient TMDL</td>
<td>Nutrient TMDL</td>
</tr>
<tr>
<td></td>
<td></td>
<td>Air temperature, flow non-linear regression</td>
</tr>
<tr>
<td>Phytoplankton</td>
<td>Set to zero</td>
<td>Set to zero</td>
</tr>
<tr>
<td></td>
<td></td>
<td>--</td>
</tr>
<tr>
<td>TSS*</td>
<td>Sediment TMDL</td>
<td>Sediment TMDL</td>
</tr>
<tr>
<td></td>
<td></td>
<td>Regression from Sediment TMDL reports</td>
</tr>
</tbody>
</table>

NOAA = National Oceanic and Atmospheric Administration; UCSD = University of California, San Diego; CalCOFI = California Cooperative Oceanic Fisheries Investigation; SCCWRP = Southern California Coastal Water Research Project.

* TSS is a constituent in the sediment transport model, which is run separately from the water quality model as discussed below.
** Refer to abbreviations in Table 5-2 in page 37.
Figure 5-1  Map of Newport Bay water quality monitoring stations (from nutrient TMDL report, County of Orange et. al, 2004).
Figure 5-2  San Diego Creek and Santa Ana Delhi Channel inflows for December 2003 - December 2006.
Figure 5-3  San Diego Creek and Santa Ana Delhi Channel inflows for December 2003 - December 2006, shown in log scale.
Figure 5-4  Santa Ana Delhi temperature: measured data (red dots), estimated (blue line) and model boundary condition (light blue line) from 2004 through 2006.

Figure 5-5  San Diego Creek temperature: measured data (red dots), estimated (blue line) and model boundary condition (light blue line) 2004 through 2006.
Figure 5-6  Santa Ana Delhi DO: measured data (red dots), estimated (blue line) and model boundary condition (light blue line) from 2004 through 2006.

Figure 5-7  San Diego Creek DO: measured data (red dots), estimated (blue line) and model boundary condition (light blue line) from 2004 through 2006.
Figure 5-8  Santa Ana Delhi NO3: measured data (red dots), estimated (blue line) and model boundary condition (light blue line) from 2004 through 2006.

Figure 5-9  San Diego Creek NO3: measured data (red dots), estimated (blue line) and model boundary condition (light blue line) from 2004 through 2006.
Figure 5-10  Santa Ana Delhi NH3: measured data (red dots), estimated (blue line) and model boundary condition (light blue line) from 2004 through 2006.

Figure 5-11  San Diego Creek NH3: measured data (red dots), estimated (blue line) and model boundary condition (light blue line) from 2004 through 2006.
Figure 5-12  Santa Ana Delhi PO4: measured data (red dots), estimated (blue line) and model boundary condition (light blue line) from 2004 through 2006.

Figure 5-13  San Diego Creek PO4: measured data (red dots), estimated (blue line) and model boundary condition (light blue line) from 2004 through 2006.
Figure 5-14  Santa Ana Delhi EC: measured data (red dots), estimated (blue line) and EC used to compute salinity model boundary condition (light blue line) 2004 through 2006.

Figure 5-15  San Diego Creek EC: measured data (red dots), estimated (blue line) and EC used to compute salinity model boundary condition (light blue line) 2004 through 2006.
Figure 5-16  Santa Ana Delhi estimated Salinity using a regression between EC and salinity from 2004 through 2006.

Figure 5-17  San Diego Creek estimated Salinity using a regression between EC and salinity from 2004 through 2006.
Figure 5-18  Regression of both water temperature and air temperature multiplied by flow rate of San Diego Creek from 2004 through 2006.

Figure 5-19  Regression of both water temperature and air temperature multiplied by flow rate of Santa Ana Delhi from 2004 through 2006.
Figure 5-20  Regression of both DO and air temperature multiplied by flow rate of Santa Ana Delhi from 2004 through 2006.

\[
\log(y) = 0.179012 \log(x)^2 + 0.142699 \log(x) + 1.208758
\]

\[R^2 = 0.826457\]

Figure 5-21  Regression of both DO and air temperature multiplied by flow rate of San Diego Creek from 2004 through 2006.

\[
\log(y) = 0.115170 \log(x)^2 + 0.336476 \log(x) + 1.120159
\]

\[R^2 = 0.760312\]
Figure 5-22  Regression of NO3 multiplied by air temperature and air temperature multiplied by flow rate of Santa Ana Delhi from 2004 through 2006.

Figure 5-23  Regression of both NO3 multiplied by air temperature and air temperature multiplied by flow rate of San Diego Creek from 2004 through 2006.
Figure 5-24  Regression of NH3 multiplied by air temperature and air temperature multiplied by flow rate of Santa Ana Delhi from 2004 through 2006.

Figure 5-25  Regression of both NH3 multiplied by air temperature and air temperature multiplied by flow rate of San Diego Creek from 2004 through 2006.
Figure 5-26  Regression of PO4 multiplied by air temperature and air temperature multiplied by flow rate of Santa Ana Delhi from 2004 through 2006.

Figure 5-27  Regression of both PO4 multiplied by air temperature and air temperature multiplied by flow rate of San Diego Creek from 2004 through 2006.
5.2.1 Development and application of deposited sediment boundary condition

Mass and distribution of deposited sediment are supplied to the water quality model as a boundary condition for the 2004 – 2006 calibration and validation period. From the sediment deposition data, the water quality model generates particulate organic matter inputs in the sediment. POM is assumed to make up 1% of the total deposited sediment (Peng et al., 2007).

5.2.1.1 Sediment transport

To generate the sediment mass and distribution boundary condition, RMA11 sediment transport simulations were run. Although water quality simulations begin in January 2004, sediment transport simulations were begun in December 2003 to capture all significant storm events that contributed sediment to the Bay during that winter. Simulations are run for the three consecutive years. However, POM is run for a full year, and then restarted from zero deposition to simulate the following year. This allowed the three years to be simulated in parallel and was justifiable because the previous year’s deposition would not have an important impact on the results. Large storms do occur near the end of 2004, however there is such a large sediment load to the Bay during early 2005 that much of the older sediment becomes buried and unavailable to the water column. Thus the impact of the load from the end of 2004 is small relative to the 2005 loads. Significant storms in the winter of 2005-2006 do not begin until around the beginning of January.

For the sediment transport simulations, hydrodynamic boundary conditions were as described in section 5.2. Daily TSS estimated data from daily flow using regressions (source: sediment TMDL reports) were used to set sediment boundary conditions for Santa Ana Delhi Channel and San Diego Creek, as shown in Figure 5-28 and Figure 5-29. Ocean boundary TSS was set to zero.

The parameters required by the sediment transport model include the number and thickness of sediment bed layers, critical shear stress for erosion for each layer, sediment density for each layer, critical shear stress for deposition, and sediment particle settling rate.

Sediment properties are a function of depth. Newly deposited sediment is generally less dense and more easily scoured than sediments which have been covered and have begun to consolidate. The variation in sediment properties is often significant over the first few centimeters of the bed. RMA11 represents the variation in bed properties as series of thin fixed layers over a variable thickness layer. Sediment is always deposited into the top layer. As the top layer fills, sediment is shifted down to the lower layers. The bottom layer can grow indefinitely. When scour occurs, the upper layers are removed first. A six-layer model was used for Newport Bay. The set of sediment properties for each layer is shown in Table 5-3. Parameters were calibrated during the Upper Newport Bay Feasibility Study for the U.S. Army Corps of Engineers (RMA, 1998).
Note that the bed layers applied in the sediment transport model are not used in the water quality model.

Deposition can only occur when the shear stress on the bed is less than the critical shear stress for deposition. The critical shear stress for deposition is only a function of the uppermost bed layer so only one value is required.

Sediment particle settling velocities are calculated in the model as the d50 (median) particle size settling velocity based on particle size distribution reported from sediment TMDL samples collected at Campus Drive (County of Orange, 2004).

Shown in Figure 5-30 through Figure 5-32 are time series plots of computed sediment deposition accumulating January 1 through December 1 of each simulation year at a location of maximum deposition in Unit I basin. These plots illustrate that virtually all of the deposition occurs during the wet season, with very little change through the dry season. The time series of sediment accumulation applied as boundary conditions in the water quality model start at zero deposition on January 1st of 2005 and 2006.

Color contour plots of deposited sediment depth on June 1 of each year are shown in Figure 5-33 through Figure 5-35. As shown in these plots and in the time series plot (Figure 5-30), sediment deposition during the winter of 2005 was far greater than during either of the other wet seasons. Sediment depth was computed assuming a density of 1000 kg/m³.

Table 5-3  Sediment model parameter values used in sediment transport simulations.

<table>
<thead>
<tr>
<th>Layer</th>
<th>Layer Thickness (cm)</th>
<th>Critical Shear Stress for Erosion (N/m²)</th>
<th>Bulk Density (kg/m³)</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>1.0</td>
<td>0.090</td>
<td>1062</td>
</tr>
<tr>
<td>2</td>
<td>2.0</td>
<td>0.120</td>
<td>1125</td>
</tr>
<tr>
<td>3</td>
<td>2.0</td>
<td>0.200</td>
<td>1187</td>
</tr>
<tr>
<td>4</td>
<td>10.0</td>
<td>0.400</td>
<td>1249</td>
</tr>
<tr>
<td>5</td>
<td>10.0</td>
<td>1.017</td>
<td>1311</td>
</tr>
<tr>
<td>6</td>
<td>--</td>
<td>2.200</td>
<td>1374</td>
</tr>
</tbody>
</table>

Critical Shear Stress for Deposition (N/m²)  0.08

*d50* Particle Size Settling Velocity (m/s)  7.9x10-5
5.2.1.2 Use of the sediment transport runs for sediment POM

The sediment runs for 2004 - 2006 were post-processed to get the daily sediment deposition or scour. Fractions of the deposited sediment are assumed for POM carbon, nitrogen and phosphorus as follows: (Peng et al., 2007; Sutula et al., 2006)

- C = 1%
- N = 0.1%
- P = 0.05%

The daily sediment deposition is used to compute the POM-C, nitrogen and phosphorus added daily to the bed according to the above percentages times the daily sediment deposition rate. The top 100 kg/m² dry weight of sediment (about 10 cm) is considered biologically active. Deposition >100 kg/m² is considered buried. Thus the max sediment related POM is as follows.

- C = 1000 g/m²
- N = 100 g/m²
- P = 50 g/m²

Any previous material in excess of these values is buried.

POM-C, nitrogen and phosphorus are lost from the bed over time by decay. POM-C consumes oxygen in the bed pore water (the bed being about 10 cm with 50% porosity) and DO diffuses from the water column to the bed depending on the difference between water column and sediment concentrations, thus more sediment deposition results in reduced oxygen concentrations in the water column. POM-N and POM-P convert to NH₃ and PO₄ in the bed via diagenesis. NH₃ and PO₄ exchange with the overlying water column. With increased sediment load resulting in more NH₃ and PO₄ in the bed, flux will tend to be from the bed to the water column, increasing water column concentrations and promoting macroalgae growth.
Figure 5-28. Santa Ana Delhi daily TSS data from 2004 through 2006.

Figure 5-29. San Diego Creek daily TSS data from 2004 through 2006.
Figure 5-30  Time series plot of simulated cumulative sediment deposition January 1 through December 31, 2004, at a location of peak deposition in Unit I basin.

Figure 5-31  Time series plot of simulated cumulative sediment deposition January 1 through December 31, 2005, at a location of peak deposition in Unit I basin.
Figure 5-32  Time series plot of simulated cumulative sediment deposition January 1 through July 1, 2006, at a location of peak deposition in Unit I basin.
Figure 5-33  Contours of simulated deposited sediment depth accumulated during the period of December 1, 2003 through June 1, 2004.
Figure 5-34 Contours of simulated deposited sediment depth accumulated during the period of January 1 through June 1, 2005.
Figure 5-35  Contours of simulated deposited sediment depth accumulated during the period of January 1 through June 1, 2006.
5.3 Initial conditions

The initial conditions for water column constituents in the Newport Bay water quality model were based on nutrient TMDL values for January of each year and set to constant values throughout the Bay as listed in Table 5-4. These initial conditions were set to give reasonable starting values, but have little impact on the modeling results as they are quickly washed away under high flow conditions. Initial conditions were set for each year to allow the years to be computed in parallel and save time. The benefit of using the ending condition from one year as the initial condition for the next year was minimal and did not justify the time cost of running the three years of simulation in series.

The initial concentrations of macroalgae were sinusoidally distributed over depth, with a maximum value of 0.1 kg/m² only on the inter-tidal region. For example, while 0.1 kg/m² of macroalgae was assigned at intertidal elevations, such as 1.5 m, 0.01 kg/m² was applied at low elevations, such as -0.6 m, declining to zero in the deepest areas. Additionally, initial macroalgae concentrations declined to zero at the higher elevations with the longest exposure time at low tide. A color contour plot of the initial conditions of macroalgae is shown in Figure 5-36.

Table 5-4 Initial conditions of water quality constituents.

<table>
<thead>
<tr>
<th>Constituents</th>
<th>DO (mg/L)</th>
<th>NH3 (mg/L)</th>
<th>NO3 (mg/L)</th>
<th>PO4 (mg/L)</th>
<th>Temperature (°C)</th>
<th>Salinity (ppt)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Values</td>
<td>8.7</td>
<td>0.089</td>
<td>0.205</td>
<td>0.029</td>
<td>13.1</td>
<td>32.3</td>
</tr>
</tbody>
</table>
Figure 5-36 Contours of the initial condition of macroalgae on January 1, 2004.
5.4 Calibration and validation results

The measured water quality data used for the Newport Bay model calibration and validation were available from nutrient TMDL reports (County of Orange et al., 2004). Data were available for the 2004 through 2006 calibration and validation period. Eight macroalgae monitoring locations are shown in Figure 5-37 (The County of Orange et al., 2004; 2005; July-September 2005; October-December 2005; January-March 2006; April-Jun 2006; July-September 2006; October-December 2006).

The simulated daily macroalgae mass summed for the whole Upper Newport Bay region is shown in Figure 5-38. The model appropriately produced the largest macroalgae concentrations during 2005. Observed data show year 2005, the wettest year during the three-year simulations period, to have the greatest macroalgae concentrations. The computed peak total macroalgae mass for 2005 was more than twice that for 2004 as shown in Figure 5-38. Computed total mass for 2004 and 2006 were somewhat similar with 2006 slightly greater.

Color contour plots of macroalgae concentration for 2004, 2005 and 2006 are provided in Figure 5-39 through Figure 5-41. While observed data showed varying comparisons between 2004 and 2006, the peaks for 2004 were generally higher than those for 2006. Timing of the measurements was different among the years, so it is difficult to say if peak concentrations were captured in all cases.

Overall, the model does an excellent job of reproducing “depth-averaged” nutrient and DO concentrations in the Bay. This is an indication that the system is well balanced in consumption and production of nutrients and DO by macroalgae growth and respiration, and sediment processes.
Figure 5-37  Map of Newport Bay macroalgae sampling stations (from nutrient TMDL report, County of Orange et. al, 2004).
Figure 5-38 Comparison of computed total macroalgae mass computed for 2004, 2005 and 2006.
Figure 5-39  Spatial contours of computed macroalgae concentration (Kg/m²) on July 1, 2004.
Figure 5-40  Spatial contours of computed macroalgae concentration (Kg/m²) on July 1, 2005.
Figure 5-41  Spatial contours of computed macroalgae concentration (Kg/m²) on July 1, 2006.
5.4.1 Calibration results in 2004

The year 2004 was a relatively dry year with corresponding smaller macroalgae blooms in the summer. The model was able to reproduce the seasonal macroalgae growth characteristics with macroalgae concentrations increasing during the spring, peaking around July and declining again in the fall. The simulated macroalgae growth also tended resurge beginning in November due to nutrient loading (except at station ALG19 where there was sloughing). Even though the model captures the seasonal trends, there were some discrepancies between the simulated macroalgae and the observed data at each station. The model tended to underpredict the peak biomass at stations including ALG24, ALG16, ALG9 and ALG7. Moreover, the model produced a more gradual buildup in macroalgae biomass, while the observed data suggest low growth until June followed by a sharp increase in biomass production. However, the aerial photography conducted three times in 2005 and the land-based photographs tend to support the more gradual buildup produced by the model. In addition, the monthly macroalgae samples are subject to great variability introduced by the sampling method as well as the naturally patchy nature of macroalgal growth (Doug Shibberu, personal communication). The model also reproduced observed spatial variation in macroalgal concentrations in the Bay. For example the model reproduced high concentrations of macroalgae observed in the Unit I (near UNBJAM) and low concentrations observed nearer the ocean. A plot comparing computed macroalgae concentration at each sampling location is shown in Figure 5-42 and plots of computed and observed macroalgae concentrations during 2004 are shown in Figure 5-43 through Figure 5-49.

Computed temperatures are in good agreement with observed data during the summer as shown in Figure 5-50 through Figure 5-53. Based on the simulation results, the spatial variation of temperature was not as significant as the temporal variation. Computed temperatures are lower than observed during September through November at all stations, as much as 4º C lower at UNBJAM, although the cause is unclear.

Plots of computed and observed dissolved oxygen are shown in Figure 5-54 through Figure 5-57. The model produces DO concentrations in the range of observed data. At the furthest downstream stations, UNBNSB and UNBCHB, the observed data seem to indicate more variation than is computed. The simulated DO did not show any anoxia for the entire year 2004 probably due to depth averaging. Near the beginning of November, in response to macroalgae growth in the model, computed DO concentrations begin to increase above the observed values. Although there are no macroalgae measurements during this time, the lack of DO response in the observed data is a likely indicator that this late fall macroalgae growth did not actually occur.

Computed and observed nutrient concentrations for 2004 including nitrate, ammonia, and phosphate are shown in Figure 5-58 through Figure 5-69. Computed nitrate concentrations correspond well with observed data, reproducing the higher values during the periods of higher inflow to the Bay, and the lower values in the summer at all the stations. The model was able to
produce the spatial variation of NO₃ (i.e., high peak concentrations at UNBJAM were between 5 and 6 mg/L while those near ocean were between 2 and 3 mg/L) indicating that NO₃ was mainly coming from San Diego Creek and that there might be sinks in the bay. Note that while computed values at the stations lower in the Bay appear to be too low, the lowest observations are at the detection limit.

Ammonia concentrations are in good agreement with observed data at stations UNBJAM and UNBSDC, slightly lower than observed and UNBNSB and even lower at UNBCHB. The model did not capture the peaks in the early of December possibly due to sediment processes which were not used in our current model. Although observed data detection limits (i.e., 0.05 and 0.1 mg/L) may be a factor for deviations between computed and observed for the lowest summer values, the downstream boundary condition may be set too low in the model. Data available for the ocean boundary were very limited and thus constant boundary condition values were used for all water quality constituents.

Phosphate concentrations are plotted in Figure 5-66 through Figure 5-69. Compute concentrations are in good agreement with observed data at all locations. At the uppermost station, UNBJAM, observed data show some greater variation than computed. This may be due to uncertainty in the creek inflow concentrations.

### 5.4.2 Validation results in 2005

The year 2005 was a wet year with relatively large macroalgae blooms in the summer. The model was able to reproduce the larger bloom relative to the other simulations years, and reproduce the seasonal macroalgae growth characteristics with macroalgae concentrations increasing during the spring, peaking around July and declining again in the fall, although with a lower peak and less rapid decline than indicated by observed data. The model continued to show a more gradual build up in macroalgal biomass compared with observations. The model reproduced observed spatial variation in macroalgal concentrations in the Bay. For example, high concentrations of macroalgae observed in the Unit I (near UNBJAM) and low concentrations observed nearer the ocean were reproduced by the model. A plot comparing computed macroalgae concentration at each sampling location is shown in Figure 5-70 and plots of computed and observed macroalgae concentrations during 2005 are shown in Figure 5-71 through Figure 5-77.

Computed temperatures are in good agreement with observed data throughout the year as shown in Figure 5-78 through Figure 5-81. Based on the simulation results, the spatial variation of temperature was not as significant as the temporal variation. Additional time series data measured at the surface and the bottom at 30 minute intervals were compared with the simulated depth averaged data at stations S1, S2 and S3 (see Figure 5-82) during June through December, shown in Figure 5-83 through Figure 5-85. Weak thermal stratification appeared in the summer (i.e., from July to September in 2005) that was not able to be captured by the 2-D model. These
data confirm that temperature is well represented in the model, although with slightly less daily variation.

Plots of computed and observed dissolved oxygen are shown in Figure 5-86 through Figure 5-89. The model produces DO concentrations in the range of observed data at station UNBSDC and slightly higher than observed at the other stations. The simulated DO tended to be higher than the observed, especially in winter (e.g., November and December), which might be related to the high concentrations of macroalgae as well as the ocean boundary condition for DO. Additional time series data available during June through December (measured at the surface and the bottom at 30 minute intervals) generally show good agreement with observed data at stations S1, S2 and S3, as shown in Figure 5-92. Measured data showed some events of anoxia in the summer and in the fall, which were not reproduced by the model.

Computed and observed nutrient concentrations including nitrate, ammonia, and phosphate for 2005 are shown in Figure 5-93 through Figure 5-104. At station UNBJAM, observed seasonal trends of nitrate concentrations are well represented by the model. At the remaining stations, computed nitrate concentrations tend to be higher than observed during storm periods and lower than observed during the summer. Observed data do not reflect the higher nitrate concentrations in the winter as they did in 2004. It is unclear whether the peaks did not actually occur or whether they were simply missed due to infrequent sampling. Some of the summertime nitrate observations are at detection limit, however, it does still appear that computed summer values are too low. The model was also able to produce the spatial variation of NO3 (i.e., peak concentrations at UNBJAM were between 6 and 7 mg/L while those near the ocean were between 2 and 3 mg/L) indicating that NO3 was mainly coming from San Diego Creek and that there might be sinks in the Bay. DO concentrations indicate that the macroalgae concentrations and associated growth and respiration are appropriate and thus should not be over-consuming nutrients, but the low computed nitrate values could be due to boundary concentrations set too low.

Ammonia concentrations are also higher than observed during the storm periods (see Figure 5-97 through Figure 5-100). Otherwise, the seasonal trends of ammonia results are in good agreement with observed data at stations UNBJAM and UNBSDC, but slightly lower than observed at UNBNSB and even lower at UNBCHB. Although observed data detection limits may be a factor for deviations between computed and observed for the lowest summer values, the downstream boundary condition may be too low due to limited data availability.

Phosphate concentrations are plotted in Figure 5-101 through Figure 5-104. Compute concentrations in general show much more seasonal variation than observed. Model results are too high in the winter and spring come into agreement with observed data during the summer, and begin to fall below observed during the fall.
5.4.3 Validation results in 2006

The year 2006 was a relatively dry year with corresponding smaller macroalgae blooms in the summer. The model was able to reproduce the seasonal macroalgae growth characteristics with macroalgae concentrations increasing during the spring, peaking around July and declining again in the fall, although with a less rapid decline than indicated by observed data. Even though the model captures the seasonal trends, there were some discrepancies between the simulated results and the observed data at each station. The model again continued to show a more gradual build up in macroalgal biomass, than the observed. The model tended to overpredict the peak biomass at almost all stations. The model produced spatial variation in macroalgal concentrations in the Bay. For example, high concentrations of macroalgae observed in the Unit I (near UNBJAM) and lower concentrations observed nearer the ocean were reproduced by the model. A plot comparing computed macroalgae concentration at each sampling location is shown in Figure 5-105 and plots of computed and observed macroalgae concentrations during 2006 are shown in Figure 5-106 through Figure 5-109.

The temporal and spatial computed temperatures are generally in good agreement with observed data as shown in Figure 5-110 through Figure 5-113. Computed temperatures are as much as 3º C lower than observed at UNBJAM during September through November. This could again be due to problem with the input meteorological data, or could be a result of inflow temperatures set too low in the model.

Plots of computed and observed dissolved oxygen are shown in Figure 5-114 through Figure 5-117. The model produces DO concentrations in March that are higher than observed by as much as 5 mg/l. This could indicate that there was macroalgae growth during the spring that is not reflected in the model, however there are no macroalgae data to confirm this. During the summer and fall, model DO at stations UNBJAM, UNBNSB, and UNBCNB is in the range of observed data but tends to fluctuate to the high side. The larger computed fluctuations may indicate that there is too much macroalgae growth in the model, and this would be confirmed by the macroalgae calibration results. At these three stations during November and December, computed values are generally higher than observed by 1 to 4 mg/L. One macroalgae data point at station ALG19 indicates there may have been a small bloom in November that was not reproduced by the model. Respiration and decay of this macroalgae could explain the lower observed DO. At the furthest downstream station, UNBCCB, the model results are in good agreement with the remaining data points during the spring and summer.

Computed and observed nutrient concentrations including nitrate, ammonia, and phosphate for 2006 are shown in Figure 5-118 through Figure 5-129. Computed nitrate concentrations correspond well with observed data, reproducing the higher values during the periods of higher inflow to the Bay, and the lower values in the summer. Note that the lowest observations are at the detection limit. The model was also able to reproduce the observed spatial variation of NO3
(i.e., peak concentrations at UNBJAM were between 5 and 6 mg/L while those near ocean were between 1 and 2 mg/L) indicating that NO3 was mainly coming from San Diego Creek and that there probably were sinks in the Bay.

Computed ammonia concentrations at stations UNBJAM, UNBSDC and UNBNSB lower than observed during the spring, late fall and winter. The higher observed values during the fall and winter may have been due to dredging the bay in 2006 that might have disturbed the biochemical system of sediments (The County of Orange et al., 2006). At UNBNSB and UNBCHB, concentrations are low during spring through mid-summer. Simulated spatial variations (e.g., peaks ranging from 0.1 – 0.5 mg/L) were much larger than the temporal variations.

Phosphate concentrations are plotted in Figure 5-126 through Figure 5-129. Compute concentrations are generally in good agreement with observed data at all locations during spring and summer but fall below observed during fall and winter. The higher observed values during the fall and winter may have been a result of the dredging. As with PO4, the simulated spatial variations (e.g., peaks ranging from 0.1 – 0.4 mg/L) were much larger than the temporal variations.

Finally, the total dry mass of macroalgae for each of the calibrated years was compared in Figure 5-130. The total dry mass in 2005 was larger than the mass in 2004 by almost two times in the summer (e.g., June, July, and August). The mass in 2004 was smaller than the mass in 2006 in the summer, which might be due to higher concentrations of nitrogen in 2006, although there were fewer storm events in 2006 than in 2004.
Figure 5-42 Comparison of computed macroalgae concentrations at sampling sites in 2004.
Figure 5-43  Comparison of computed macroalgae concentrations with the measured data at ALG24 in 2004.

Figure 5-44  Comparison of computed macroalgae concentrations with the measured data at ALG19 in 2004.
Figure 5-45  Comparison of computed macroalgae concentrations with the measured data at ALG16 in 2004.

Figure 5-46  Comparison of computed macroalgae concentrations with the measured data at ALG9 in 2004.
Figure 5-47  Comparison of computed macroalgae concentrations with the measured data at ALG7 in 2004.

Figure 5-48  Comparison of computed macroalgae concentrations with the measured data at ALG4 in 2004.
Figure 5-49 Comparison of computed macroalgae concentrations with the measured data at ALG2 in 2004.
Figure 5-50  Comparison of computed temperature (°C) with the measured data at UNBJAM in 2004.

Figure 5-51  Comparison of computed temperature (°C) with the measured data at UNBSDC in 2004.
Figure 5-52  Comparison of computed temperature (°C) with the measured data at UNBNSB in 2004.

Figure 5-53  Comparison of computed temperature (°C) with the measured data at UNBCHB in 2004.
Figure 5-54 Comparison of computed DO concentrations with the measured data at UNBJAM in 2004.

Figure 5-55 Comparison of computed DO concentrations with the measured data at UNBSDC in 2004.
Figure 5-56  Comparison of computed DO concentrations with the measured data at UNBNSB in 2004.

Figure 5-57  Comparison of computed DO concentrations with the measured data at UNBCHB in 2004.
Figure 5-58  Comparison of computed NO3 concentrations with the measured data at UNBJAM in 2004.

Figure 5-59  Comparison of computed NO3 concentrations with the measured data at UNBSDC in 2004.
Figure 5.60  Comparison of computed NO3 concentrations with the measured data at UNBNSB in 2004.

Figure 5.61  Comparison of computed NO3 concentrations with the measured data at UNBCHB in 2004.
Figure 5-62  Comparison of computed NH3 concentrations with the measured data at UNBJAM in 2004.

Figure 5-63  Comparison of computed NH3 concentrations with the measured data at UNBSDC in 2004.
Figure 5-64  Comparison of computed NH3 concentrations with the measured data at UNBNSB in 2004.

Figure 5-65  Comparison of computed NH3 concentrations with the measured data at UNBCHB in 2004.
Figure 5-66 Comparison of computed PO4 concentrations with the measured data at UNBJAM in 2004.

Figure 5-67 Comparison of computed PO4 concentrations with the measured data at UNBSDC in 2004.
Figure 5-68  Comparison of computed PO4 concentrations with the measured data at UNBNSB in 2004.

Figure 5-69  Comparison of computed PO4 concentrations with the measured data at UNBCHB in 2004.
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Figure 5-71  Comparison of computed macroalgae concentrations with the measured data at ALG24 in 2005.

Figure 5-72  Comparison of computed macroalgae concentrations with the measured data at ALG19 in 2005.
Figure 5-73  Comparison of computed macroalgae concentrations with the measured data at ALG16 in 2005.

Figure 5-74  Comparison of computed macroalgae concentrations with the measured data at ALG9 in 2005.
Figure 5-75  Comparison of computed macroalgae concentrations with the measured data at ALG7 in 2005.

Figure 5-76  Comparison of computed macroalgae concentrations with the measured data at ALG4 in 2005.
Figure 5-77  Comparison of computed macroalgae concentrations with the measured data at ALG2 in 2005.
Figure 5-78  Comparison of computed temperature (°C) with the measured data at UNBJAM in 2005.

Figure 5-79  Comparison of computed temperature (°C) with the measured data at UNBSDC in 2005.
Figure 5-80  Comparison of computed temperature (°C) with the measured data at UNBNSB in 2005.

Figure 5-81  Comparison of computed temperature (°C) with the measured data at UNBCHB in 2005.
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Figure 5-94  Comparison of computed NO3 concentrations with the measured data at UNBSDC in 2005.
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Figure 5-96 Comparison of computed NO3 concentrations with the measured data at UNBCHB in 2005.
Figure 5-97  Comparison of computed NH3 concentrations with the measured data at UNBJAM in 2005.

Figure 5-98  Comparison of computed NH3 concentrations with the measured data at UNBSDC in 2005.
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Figure 5-100  Comparison of computed NH3 concentrations with the measured data at UNBCHB in 2005.
Figure 5-101  Comparison of computed PO4 concentrations with the measured data at UNBJAM in 2005.

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Figure 5-106  Comparison of computed macroalgae concentrations with the measured data at ALG24 in 2006.

Figure 5-107  Comparison of computed macroalgae concentrations with the measured data at ALG19 in 2006.
Figure 5-108  Comparison of computed macroalgae concentrations with the measured data at ALG16 in 2006.

Figure 5-109  Comparison of computed macroalgae concentrations with the measured data at ALG7 in 2006.
Figure 5-110  Comparison of computed temperature (°C) with the measured data at UNBJAM in 2006.

Figure 5-111  Comparison of computed temperature (°C) with the measured data at UNBSDC in 2006.
Figure 5-112  Comparison of computed temperature (°C) with the measured data at UNBNSB in 2006.

Figure 5-113  Comparison of computed temperature (°C) with the measured data at UNBCHB in 2006.
Figure 5-114  Comparison of computed DO concentrations with the measured data at UNBJAM in 2006.

Figure 5-115  Comparison of computed DO concentrations with the measured data at UNBSDC in 2006.
Figure 5-116  Comparison of computed DO concentrations with the measured data at UNBNSB in 2006.

Figure 5-117  Comparison of computed DO concentrations with the measured data at UNBCHB in 2006.
Figure 5-118  Comparison of computed NO3 concentrations with the measured data at UNBJAM in 2006.

Figure 5-119  Comparison of computed NO3 concentrations with the measured data at UNBSDC in 2006.
Figure 5-120 Comparison of computed NO3 concentrations with the measured data at UNBNSB in 2006.

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Figure 5-122  Comparison of computed NH3 concentrations with the measured data at UNBJAM in 2006.

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Figure 5-124  Comparison of computed NH3 concentrations with the measured data at UNBNSB in 2006.

Figure 5-125  Comparison of computed NH3 concentrations with the measured data at UNBCHB in 2006.
Figure 5-126  Comparison of computed PO4 concentrations with the measured data at UNBJAM in 2006.

Figure 5-127  Comparison of computed PO4 concentrations with the measured data at UNBSDC in 2006.
Figure 5-128 Comparison of computed PO4 concentrations with the measured data at UNBNSB in 2006.

Figure 5-129 Comparison of computed PO4 concentrations with the measured data at UNBCHB in 2006.
Figure 5-130 Comparison of computed total dry mass of macroalgae for calibration years of 2004, 2005 and 2006.
5.5 Comparison of Nutrient concentrations in the sediments

The model was not calibrated to measured porewater concentrations, however a post-calibration comparison with observed data shows reasonably good agreement. Sediment porewater samples of ammonia and phosphate were collected at the stations shown in Figure 5-131 during 2004 (Sutula, et al., 2004). Comparison plots between the collected data (at 0 to 1 cm sediment depth) and simulated porewater concentrations (from active sediment layer of 0 cm up to a maximum of approximately 10 cm sediment depth) are provided in Figure 5-132 through Figure 5-143. These plots show that the model computes reasonable porewater concentrations of ammonia and phosphate.
Figure 5-131 Map of sediment porewater sampling sites in Upper Newport Bay (UNB) (Source: Sutula et al., 2006).
Figure 5-132  Comparison of surface sediment observed (0-1 cm) and computed (up to 10 cm) NH3 at the intertidal region of NB1 (Site 1 in Figure 5-131).

Figure 5-133  Comparison of surface sediment observed (0-1 cm) and computed (up to 10 cm) PO4 at the intertidal region of NB1 (Site 1 in Figure 5-131).
Figure 5-134  Comparison of surface sediment observed (0-1 cm) and computed (up to 10 cm) NH3 at the subtidal region of NB1 (Site 1 in Figure 5-131).

Figure 5-135  Comparison of surface sediment observed (0-1 cm) and computed (up to 10 cm) PO4 at the subtidal region of NB1 (Site 1 in Figure 5-131).
Figure 5-136  Comparison of surface sediment observed (0-1 cm) and computed (up to 10 cm) NH3 at the intertidal region of NB2 (Site 2 in Figure 5-131).

Figure 5-137  Comparison of surface sediment observed (0-1 cm) and computed (up to 10 cm) PO4 at the intertidal region of NB2 (Site 2 in Figure 5-131).
Figure 5-138 Comparison of surface sediment observed (0-1 cm) and computed (up to 10 cm) NH3 at the subtidal region of NB2 (Site 2 in Figure 5-131).

Figure 5-139 Comparison of surface sediment observed (0-1 cm) and computed (up to 10 cm) PO4 at the subtidal region of NB2 (Site 2 in Figure 5-131).
Figure 5-140  Comparison of surface sediment observed (0-1 cm) and computed (up to 10 cm) NH3 at the intertidal region of NB3 (Site 3 in Figure 5-131).

Figure 5-141  Comparison of surface sediment observed (0-1 cm) and computed (up to 10 cm) PO4 at the intertidal region of NB3 (Site 3 in Figure 5-131).
Figure 5-142 Comparison of surface sediment observed (0-1 cm) and computed (up to 10 cm) NH3 at the subtidal region of NB3 (Site 3 in Figure 5-131).

Figure 5-143 Comparison of surface sediment observed (0-1 cm) and computed (up to 10 cm) PO4 at the subtidal region of NB3 (Site 3 in Figure 5-131).
5.6 Limiting factors for macroalgae growth

Four of the limiting factors of macroalgae growth (light, internal nitrogen, phosphorus and freshwater stress) are plotted for each of the monitoring stations and for each year during the calibration/validation period in Figure 5-144 through Figure 5-147. These plots show the relative importance of each of the four factors. Note that these are not the only limitation factors in the model formulation. For example temperature limitation, not included in these plots, becomes of primary importance during the winter.

5.6.1 Limiting factors in 2004

Four limiting factors of macroalgae growth in 2004 are plotted for three macroalgae monitoring stations (i.e., head (Station ALG19), mid (Station ALG13) and lower part (Station ALG7) of the Bay) in Figure 5-144. These plots indicate that nitrogen is the primary limiting factor among the four during March through early June and November through December. Nitrogen has been considered as limiting nutrient in eutrophic coastal marine ecosystems (Howarth and Marino, 2006). During February and June through early August, nitrogen and phosphorus limitation are alternately important and dependent on location. During January and mid-August through mid-October, phosphorus limitation dominates. Phosphorus limitation occurs more often at locations lower in the Bay. Kamer et al. (2002) also found that nitrogen as the primary limiting nutrient with phosphorus being secondary in the Newport Bay. Even though none of the experiments produced biomass increases with phosphorus addition only, addition of phosphorus stimulated growth when supplied with nitrogen. Freshwater stress becomes important during storm periods. Light is limiting every night but seldom during the day.

5.6.2 Limiting factors in 2005

Four limiting factors of macroalgae growth in 2005 are plotted for three macroalgae monitoring stations (i.e., head, mid and lower part of the Bay) in Figure 5-145. These plots indicate that nitrogen is the primary limiting factor among the four during January through July except during storm periods when freshwater stress dominates. During August, nitrogen and phosphorus limitation are alternately important with phosphorus limitation occurring primarily at the uppermost stations. By September through the end of the year phosphorus limitation is the primary day time limitation factor. Light is limiting every night but seldom during the day.

The massive extent of the bloom in 2005 suggests that there was no significant nutrient limitation. Almost all of the available inter-tidal region was occupied by dense macroalgal mats during the peak bloom periods in summer and fall. To take a closer look at this period, Figure 5-146 shows a plot of the actual values of light, internal nitrogen, phosphorus, space and temperature limitation at the head of the Bay (ALG 19). What this plots show is that in this area of most dense macroalgal growth, the space limitation factor (not included in Figure 5-145) becomes dominant during July and August. Space limitation only becomes dominant at the head of the Bay during the summer of 2005, and not all elsewhere in the Bay, or during 2004 or 2006.
5.6.3 Limiting factors in 2006

Four limiting factors of macroalgae growth in 2006 are plotted for three macroalgae monitoring stations (i.e., head, mid and lower part of the Bay) in Figure 5-147. These plots indicate that phosphorus is the primary limiting factor among the four throughout the year at the stations lower in Newport Bay. At these stations there are some brief periods of nitrogen limitation during January, March and April. The uppermost stations are also primarily dominated by phosphorus limitation, however during March through mid-July nitrogen and phosphorus limitation are alternately important. Freshwater stress becomes important during storm periods. Light is limiting every night but seldom during the day.

This is a confounding result as nitrogen concentrations/loads in 2006 were reduced as compared to 2005. So there must have been a proportionally greater decline in phosphorus availability.
Figure 5-144  Limiting factors of macroalgae growth at each monitoring stations in 2004. The values of one, two, three and four indicate light limiting, internal nitrogen, phosphorus and freshwater stress, respectively. (a) Station ALG19; (b) Station ALG13; (c) Station ALG7.
Figure 5-145  Limiting factors of macroalgae growth at each monitoring stations in 2005. The values of one, two, three and four indicate light limiting, internal nitrogen, phosphorus and freshwater stress, respectively. (a) Station ALG19; (b) Station ALG13; (c) Station ALG7.
Figure 5-146 Time series plot of light, internal nitrogen, phosphorus, space and temperature limitation factors at Station ALG19 during summer 2005.
Figure 5-147 Limiting factors of macroalgae growth at each monitoring stations in 2006. The values of one, two, three and four indicate light limiting, internal nitrogen, phosphorus and freshwater stress, respectively. (a) Station ALG19; (b) Station ALG13; (c) Station ALG7.
5.7 Nutrient cycles in Newport Bay: Discussion of nutrient budgets

Nutrient budgets of nitrogen and phosphorus were estimated based on model simulations from 2004 through 2006. The regional area for the budgets is the same as that of the model simulation including from Unit 1 Basin to harbor entrance. The budgets consider two inflows including Santa Ana Delhi Channel and San Diego Creek, and a tidal outflow at the harbor entrance. The model was based on calendar years, therefore loads generated by the stormy winter of 2004-2005 are split between 2004 and 2005, assuming that late fall storms do not contribute significantly to nutrient supply for the following summer.

5.7.1 Nitrogen

The nitrogen budget consists of two inflows as sources, and one outflow and denitrification as sinks. The differences between sources and sinks for each year are the net changes in nitrogen in the water column and sediments plus nitrogen that has been taken up by macroalgae, which either remains in the Bay or has been removed from the system. The total inflows, outflows, denitrification and net changes for each year are listed in Table 5-5.

Nitrogen-containing suspended sediments coming from inflows are deposited on the bed. These deposited organic sediments are decayed and release dissolved nitrogen into the water, adding to the water column dissolved nitrogen from the creek flows. The cumulative contributions of total nitrogen to the system for each of the three calibration years are shown as time series in Figure 5-148, Figure 5-150 and Figure 5-152.

Time series of nitrogen distribution in the Bay are plotted for each year in Figure 5-149, Figure 5-151 and Figure 5-153, respectively. As nitrogen enters the system, these plots show how it is distributed in the Bay. Any buried sediment is cumulative because once it is buried it is permanently lost from the system. The active sediment layer increases when storm inflows deposit sediment, and decreases as nitrogen is released to the water column. Sloughing and non-recycled dead macroalgae is cumulative because nitrogen is permanently lost from the system. Nitrogen in the macroalgae rises and falls based on macroalgae growth dynamics. Water column concentrations vary, with the largest changes resulting from increased inflows. Nitrogen that exits the ocean boundary and denitrification are both cumulative as they are permanent losses from the system.

The final cumulative total of nitrogen at the end of each year in the distribution plots is equal to the end of year total of nitrogen in the inflows for each year in Figure 5-148, Figure 5-150 and Figure 5-152. Looking back at the net changes in Table 5-5, the net change for each year is equal to the sum of the nitrogen in the buried sediment, active sediment layer, macroalgae, sloughed and non-recycled dead macroalgae and the water column.
5.7.1.1 Sources

- **Inflows**
  Daily discharges of Santa Ana Delhi Channel (SADF01) and San Diego Creek at Campus Drive (SDMF05) were multiplied by their nitrogen constituent concentrations. These concentrations are the same as the inflow inputs. The amount of total nitrogen of inflows is shown in Table 5-5.

- **Minor sources**
  Potential sources of unknown magnitude include groundwater and non-point sources. Groundwater can be considered negligible at less than 24 m$^3$/day (Worsnop et al). In addition, nitrogen data from wells adjacent to the bay show nitrogen concentrations average less than 1 mg/L (personal communication with Santa Ana Regional Water Quality Control Board). Combining these two sources of information, the max load would be less than 10 kg/yr (22 lbs/yr). Other minor sources that have been neglected include small drainages – Bonita Canyon, Costa Mesa Channel, Santa Isabella Channel, Big Canyon, and seepage from bluffs at south end of bay (e.g., for Costa Mesa Channel is 259 kg/yr (572 lbs/yr) TN dry season average, 389 kg/yr (858 lbs/yr) wet season average). In the nitrogen budget, minor sources are not considered.

  Tidal mixing is a source of nitrogen in the Bay, however this is difficult to quantify with available data.

5.7.1.2 Sinks

- **Outflow**
  Daily discharge from Newport Bay was calculated in the model. This discharge was multiplied by the computed nitrogen constituent concentrations at the tidal boundary. The amount of total nitrogen in the outflow is about 299,506,215 lbs/yr.

- **Denitrification**
  Denitrification of nitrate to N$_2$ is estimated in the model as previously described in section 3.2. The amount of total nitrogen lost from the system due to denitrification is about 40,045,021 lbs/yr. It is interesting that the denitrification rate didn’t increase greatly in 2005, which could be explained by the efficient uptake of nitrate by the large macroalgal bloom that year. This likely minimized the pool of nitrate available for denitrification. Low denitrification rates were also observed in the intertidal region of the Bay (personal communication with Peggy Fong and Tonya Kane).
5.7.2 Phosphorus

The phosphorus budget consists of two sources (i.e., two inflows) and a sink (one outflow) and includes phosphate and particulate phosphorus. As with the nitrogen budget, the differences between sources and sinks for each year are the net changes in phosphorus in the water column and sediments plus phosphorus that has been taken up by macroalgae, which either remains in the Bay or has been removed from the system. The total inflows, outflows and net changes for each year are listed in Table 5-6.

Phosphorus-containing suspended sediments coming from inflows are deposited on the bed. These deposited organic sediments are decayed and release dissolved phosphorus into the water, adding to water column dissolved phosphorus from the creek flows. The cumulative contributions of phosphorus to the system for each of the three calibration years are shown as time series in Figure 5-154, Figure 5-156 and Figure 5-158.

Time series of phosphorus distribution in the Bay are plotted for each year in Figure 5-155, Figure 5-157 and Figure 5-159, respectively. As phosphorus enters the system, these plots show how it is distributed in the Bay. Any buried sediment is cumulative because once it is buried it is permanently lost from the system. The active sediment layer increases when storm inflows deposit sediment, and decreases as phosphorus is released to the water column. Sloughing and non-recycled dead macroalgae is cumulative because phosphorus is permanently lost from the system. Phosphorus in the macroalgae rises and falls based on macroalgae growth dynamics. Water column concentrations vary, with the largest changes resulting from increased inflows. Phosphorus that exits the ocean boundary is cumulative as it is a permanent loss from the system.

The final cumulative total of phosphorus at the end of each year in the distribution plots is equal to the end of year total of phosphorus in the inflows for each year in Figure 5-154, Figure 5-156 and Figure 5-158. Looking back at the net changes in Table 5-6, the net change for each year is equal to the sum of the phosphorus in the buried sediment, active sediment layer, macroalgae, sloughed and non-recycled dead macroalgae and the water column.

5.7.2.1 Sources

- **Inflows**

  Daily discharges of Santa Ana Delhi Channel (SADF01) and San Diego Creek at Campus Drive (SDMF05) were multiplied by their phosphorus constituent concentrations. The amount of total nitrogen of inflows is shown in Table 5-6.

  Tidal mixing is an important source of phosphorus in the Bay, however this is difficult to quantify with available data.
5.7.2.2 Sinks

- **Outflow**

Daily discharge from Newport Bay was calculated in the model. This discharge was multiplied by the computed phosphorus constituent concentrations. The amount of total phosphorus of outflows is shown in Table 5-6.

<table>
<thead>
<tr>
<th>Table 5-5  Newport Bay nitrogen budget for calibration years.</th>
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<td></td>
<td>Inflows (lbs)</td>
<td>Outflow (lbs)</td>
<td>Denitrification (lbs)</td>
<td>Net changes (lbs)</td>
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<tr>
<td>Jan, 2004 – Dec, 2004</td>
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<td>319,000</td>
<td>63,000</td>
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<td>Jan, 2005 – Dec, 2005</td>
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<td>538,000</td>
<td>71,000</td>
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<table>
<thead>
<tr>
<th>Table 5-6  Newport Bay phosphorus budget for calibration years.</th>
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<tbody>
<tr>
<td></td>
<td>Inflows (lbs)</td>
<td>Outflow (lbs)</td>
<td>Net changes (lbs)</td>
</tr>
<tr>
<td>Jan, 2004 – Dec, 2004</td>
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<td>65,000</td>
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<tr>
<td>Jan, 2005 – Dec, 2005</td>
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<td>169,000</td>
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</tr>
<tr>
<td>Jan, 2006 – Dec, 2006</td>
<td>15,000</td>
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</tr>
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</table>
Figure 5-148. Time series of accumulated nitrogen mass into Newport Bay from January, 2004 to December, 2004 as input to the model. Very dark blue indicates particulate nitrogen from two inflows, cyan and light blue indicate NH3 and NO3 of San Diego Creek, respectively, and dark blue and blue indicate NH3 and NO3 of Santa Ana Delhi.
Figure 5-149. Time series of distribution of nitrogen in Newport Bay as computed in the model from January, 2004 to December, 2004. Dark brown indicates buried sediments, light brown indicates active sediment layer including N-POM, dark green indicates sloughing and non-recycled dead macroalgae (NRM), light green indicates internal nitrogen, light blue indicates NH3 in the water column, cyan indicates NO3 in the water column, very dark blue indicates particulate nitrogen out to the ocean, dark blue indicates NH3 out to the ocean, blue indicates NO3 out to the ocean and pale blue indicates denitrification.
Figure 5-150. Time series of accumulated nitrogen mass into Newport Bay from January, 2005 to December, 2005 as input to the model. Very dark blue indicates nitrogen from two inflows, cyan and light blue indicate NH3 and NO3 of San Diego Creek, respectively, and dark blue and blue indicate NH3 and NO3 of Santa Ana Delhi.
Figure 5-151. Time series of distribution of nitrogen in Newport Bay as computed in the model from January, 2005 to December, 2005. Dark brown indicates buried sediments, light brown indicates active sediment layer including N-POM, dark green indicates sloughing and non-recycled dead macroalgae (NRM), light green indicates internal nitrogen, light blue indicates NH3 in the water column, cyan indicates NO3 in the water column, very dark blue indicates particulate nitrogen out to the ocean, dark blue indicates NH3 out to the ocean, blue indicates NO3 out to the ocean and pale blue indicates denitrification.
Figure 5-152. Time series of accumulated nitrogen mass into Newport Bay from January, 2006 to June, 2006 as input to the model. Very dark blue indicates nitrogen from two inflows, cyan and light blue indicate NH$_3$ and NO$_3$ of San Diego Creek, respectively, and dark blue and blue indicate NH$_3$ and NO$_3$ of Santa Ana Delhi.
Figure 5-153. Time series of distribution of nitrogen in Newport Bay as computed in the model from January, 2006 to June, 2006. Dark brown indicates buried sediments, light brown indicates active sediment layer including N-POM, dark green indicates sloughing and non-recycled dead macroalgae (NRM), light green indicates internal nitrogen, light blue indicates NH3 in the water column, cyan indicates NO3 in the water column, very dark blue indicates particulate nitrogen out to the ocean, dark blue indicates NH3 out to the ocean, blue indicates NO3 out to the ocean and pale blue indicates denitrification.
Figure 5-154. Time series of distribution of phosphorus in Newport Bay as computed in the model from January, 2004 to December, 2004. Dark brown indicates buried sediments, light brown indicates active sediment layer including P-POM, dark green indicates sloughing and non-recycled dead macroalgae (NRM), light green indicates phosphorus in macroalgae, cyan indicates PO4 in the water column, very dark blue indicates particulate phosphorus out to the ocean, and blue indicates PO4 out to the ocean.
Figure 5-155. Time series of accumulated phosphorus mass into Newport Bay from January, 2004 to December, 2004 as input to the model. Very dark blue indicates particulate phosphorus from two inflows, cyan and light blue indicate NH3 and NO3 of San Diego Creek, respectively, and dark blue and blue indicate NH3 and NO3 of Santa Ana Delhi.
Figure 5-156. Time series of accumulated phosphorus mass into Newport Bay from January, 2005 to December, 2005 as input to the model. Very dark blue indicates particulate phosphorus from two inflows, cyan and light blue indicate NH3 and NO3 of San Diego Creek, respectively, and dark blue and blue indicate NH3 and NO3 of Santa Ana Delhi.
Figure 5-157. Time series of distribution of phosphorus in Newport Bay as computed in the model from January, 2005 to December, 2005. Dark brown indicates buried sediments, light brown indicates active sediment layer including P-POM, dark green indicates sloughing and non-recycled dead macroalgae (NRM), light green indicates phosphorus in macroalgae, cyan indicates PO4 in the water column, very dark blue indicates particulate phosphorus out to the ocean, and blue indicates PO4 out to the ocean.
Figure 5-158. Time series of accumulated phosphorus mass into Newport Bay from January, 2006 to June, 2006 as input to the model. Very dark blue indicates particulate phosphorus from two inflows, cyan and light blue indicate NH3 and NO3 of San Diego Creek, respectively, and dark blue and blue indicate NH3 and NO3 of Santa Ana Delhi.
Figure 5-159. Time series of distribution of phosphorus in Newport Bay as computed in the model from January, 2006 to June, 2006. Dark brown indicates buried sediments, light brown indicates active sediment layer including P-POM, dark green indicates sloughing and non-recycled dead macroalgae (NRM), light green indicates phosphorus in macroalgae, cyan indicates PO4 in the water column, very dark blue indicates particulate phosphorus out to the ocean, and blue indicates PO4 out to the ocean.
5.7.3 Nitrate loading during a storm period

To take a closer look at the fate of nitrate loading during storm periods, the two-month wet weather period of January – February 2004 is analyzed. Plots are provided in Figure 5-160 that show nitrate inflow, outflow and water column concentration. A time series of creek inflow is provided to show corresponding flow conditions. These plots show a rapid increase in water column nitrate concentrations during storm events. The nitrate passes quickly through the Bay and discharges to the ocean allowing water column concentrations to return to pre-storm levels. Thus it appears that the storm-related nitrate load has little long-term impact on the system.
Figure 5-160  NO3-N for storm events between February and March 2004.
6. Additional model simulations, results and discussion

6.1 Scenarios with conservative constituents

Model simulations were run with nutrients (nitrate, ammonia and phosphate) simulated as conservative constituents. There was no interaction among the nutrients, or with macroalgae or other constituents, and heat budget was not considered. Two simulations were run for each constituent: one with boundary conditions set as described in section 4.1.2; and one with the same creek inflow boundaries, but tidal boundary concentrations set to zero. Comparison of results from these two types of simulations and comparisons between simulation results and observed data can help determine the relative importance of the tidal and creek inflow boundary contributions to total nutrient concentrations in the Bay.

Conservatively computed NO3 is plotted with observed data in Figure 6-1 through Figure 6-4 at stations shown in Figure 5-1. In these simulations, the impact of the ocean boundary condition is least important at the most upstream station (UNBJAM) and becomes more important at the furthest downstream station (UNBCHB) as expected. At UNBJAM and UNBSDC, computed concentrations are higher than observed much of the time, although computed results correspond well with several data points. At UNBNSB, the simulation results correspond well with much of the observed data. At UNBCHB, the observed data points fall in between the two simulations during the dry season and the wet season results vary by year.

These conservative NO3 simulation results indicate that the ocean boundary concentrations do not dominate the system. Compared with the observed data, it also appears that there is a sink of NO3 in the system, as predicted from the nitrogen mass balance discussed in section 5.7.1. Macroalgal uptake and/or sediments (i.e., probably via denitrification) are likely to be the primary sink for NO3.

Conservatively computed NH3 is plotted with observed data in Figure 6-5 through Figure 6-8 at stations shown in Figure 5-1. In these simulations, the impact of the ocean boundary condition is fairly important throughout the Bay, but particularly further downstream. Even with concentrations set for the ocean boundary, simulation results are lower than observed data, with very few exceptions, at all stations. Compared with the observed data, it is apparent that there is a source of NH3 in the system, which increases the importance of the NO3 sink, given that the nitrogen mass balance discussed in section 5.7.1 shows a net loss of nitrogen in the system. Sediments (e.g., diagenesis of POM) are likely to be the primary source of NH3.

Conservatively computed PO4 is plotted with observed data in Figure 6-9 through Figure 6-12 at stations shown in Figure 5-1. In these simulations, the impact of the ocean boundary condition is
quite important throughout the Bay, to the point that ocean input alone can produce summertime concentrations at UNBSDC that are in the range of observed values and, and concentrations even higher than observed at the further downstream stations. An internal source of PO4 is indicated at UNBJAM, given that computed concentrations are lower than observed. This is in line with the predictions in the mass balance. Sediments (e.g., diagenesis of POM) are the likely source, especially in summer. At the stations further downstream, however, a net sink is indicated. Although there are both sources and sinks of all nutrients in the system, another likely explanation is that the tidal boundary is actually set too high, thus causing the elevated concentrations lower in the Bay. Nutrient data for the ocean boundary are very limited (collected quarterly by from a different location - LA Harbor) and therefore add a higher level of uncertainty, which could be an important factor affecting the ability to closely calibrate the model. Moreover, upwelling events in the west coast of the Pacific Ocean (which are not always captured in the ocean nutrient data) potentially add uncertainty of tidal boundary conditions to the nutrient cycles of the Bay, since the driven surface waters from the coast are replaced by denser, cold water below carrying dissolved nutrients. The use of lower Bay phosphate data as the ocean boundary condition may have improved the results.
Figure 6-1  Observed and conservatively computed NO3 at UNBJAM for years 2004 - 2006. Conservative simulations were performed both with tidal boundaries set using observed data, and set to zero.
Figure 6-2  Observed and conservatively computed NO3 at UNBSDC for years 2004 - 2006. Conservative simulations were performed both with tidal boundaries set using observed data, and set to zero.
Figure 6-3  Observed and conservatively computed NO3 at UNBNSB for years 2004 - 2006. Conservative simulations were performed both with tidal boundaries set using observed data, and set to zero.
Figure 6-4  Observed and conservatively computed NO3 at UNBCHB for years 2004 - 2006. Conservative simulations were performed both with tidal boundaries set using observed data, and set to zero.
Figure 6-5  Observed and conservatively computed NH3 at UNBJAM for years 2004 - 2006. Conservative simulations were performed both with tidal boundaries set using observed data, and set to zero.
Figure 6-6  Observed and conservatively computed NH3 at UNBSDC for years 2004 - 2006. Conservative simulations were performed both with tidal boundaries set using observed data, and set to zero.
Figure 6-7  Observed and conservatively computed NH3 at UNBNSB for years 2004 - 2006.  Conservative simulations were performed both with tidal boundaries set using observed data, and set to zero.
Figure 6-8  Observed and conservatively computed NH3 at UNBCHB for years 2004 - 2006. Conservative simulations were performed both with tidal boundaries set using observed data, and set to zero.
Figure 6-9  Observed and conservatively computed PO4 at UNBJAM for years 2004 - 2006. Conservative simulations were performed both with tidal boundaries set using observed data, and set to zero.
Figure 6-10  Observed and conservatively computed PO4 at UNBSDC for years 2004 - 2006. Conservative simulations were performed both with tidal boundaries set using observed data, and set to zero.
Figure 6-11  Observed and conservatively computed PO4 at UNBNSB for years 2004 - 2006. Conservative simulations were performed both with tidal boundaries set using observed data, and set to zero.
Figure 6-12  Observed and conservatively computed PO4 at UNBCHB for years 2004 - 2006. Conservative simulations were performed both with tidal boundaries set using observed data, and set to zero.
6.2 Scenarios with biological and chemical reactions

To analyze the relative importance of sediment processes, simulations were run with biological and chemical reactions but no sediment interactions. All other model parameters were set the same as for the calibrated model.

In Figure 6-13 through Figure 6-15 total dry weight macroalgae, computed with no sediment interactions, is compared with the calibrated model result (with sediment interactions) for years 2004 through 2006. In 2004, macroalgae mass declines continually from the initial condition on January first, nearing zero by September. In 2005 and 2006, peak macroalgae mass occurs in the spring at very small fractions of the peaks in the calibrated model, then declines to minimal values by the end of the year.

These results give strong evidence that sediments provide an important source of nutrients for macroalgae blooms in Newport Bay.

The relative importance of NO3 inputs from two main inflows was also analyzed. Simulations were run under the same conditions as the calibration but with no NO3 coming from inflows. In Figure 7-16, Figure 7-17 and Figure 7-18, total dry weight of macroalgae simulated for each calibration year is plotted with and without NO3 inputs from inflows. The model simulations indicate that removal of NO3 from inflows could decrease total dry weight by 30% to 50% of the peak biomass in a year. It may be presumed that an ideal approach to manage macroalgal blooms would be complete removal of NO3 from the inflows. However, these simulation results in comparison with results from the simulations without sediments indicate that control of particulate nitrogen could have a more significant impact than NO3 reductions on management of macroalgal blooms.
Figure 6-13  Comparison of total dry mass of macroalgae simulated with and without sediment fluxes in 2004.
Figure 6-14 Comparison of total dry mass of macroalgae simulated with and without sediment fluxes in 2005.
Figure 6-15  Comparison of total dry mass of macroalgae simulated with and without sediment fluxes in 2006.
Figure 6-16  Comparison of total dry mass of macroalgae simulated with and without NO3 inputs from inflows in 2004.
Figure 6-17  Comparison of total dry mass of macroalgae simulated with and without NO3 inputs from inflows in 2005.
Figure 6-18  Comparison of total dry mass of macroalgae simulated with and without NO3 inputs from inflows in 2006.
6.3 Residence time simulations

Residence time simulations were performed for the dry weather period of July – August 2004. To compute residence time, all flows entering at the creek inflow or tidal boundaries are considered “new” water with a residence time of zero. The water volume in the 2-D area of the model is loaded with a tracer at 1 g/m3-day. The computed tracer concentration in the Bay is a direct measure of residence time (in days) of water within that area.

The Bay is initialized with zero tracer concentration throughout. Over time, tracer concentration – residence time – approaches an equilibrium concentration that varies with spring-neap tidal variation. During the summer, when tributary inflows are low, residence time increases through the neap tide and decreases through the spring tide. Lowest residence times follow the strongest spring tides. Color contour plots of residence time in Figure 6-19 bracket the range in residence time through the spring-neap tidal cycle. Figure 6-20 shows the timing of these plots relative to the tidal cycle. The highest residence time, about 12 days, occurs in Rhine Channel after the neap tide.
Figure 6-19  Color contour plots showing the range of residence times occurring during the 01 July – 31 August 2004 simulation period. Tidal phase and stage at the times of these plots can be seen in Figure 6-20.
Figure 6-20  Time series of stage applied at the downstream model boundary for the 01 July – 31 August 2004 residence time simulation. Red dots indicate times of contour plots in Figure 6-19.
7. Reduced nutrient load simulations

To aid the Regional Board staff in selection of appropriate numeric water quality objectives for nutrients entering Newport Bay, the calibrated model has been used to simulate macroalgae blooms for reduced loading alternatives.

The approach was to limit total nitrogen entering the Bay based on concentration limits for the inflows. Different limits were set for storm flows versus low flow periods. A storm flow was considered to be any combined inflow (San Diego Creek plus Santa Ana Delhi Channel) greater than 50 cfs. Two different limits were set for storm flows (3 and 5 mg/L) and three for low flow periods (0.5, 1 and 2 mg/L), and each combination was simulated. For each alternative, concentrations of NH3 and NO3 of each inflow were allotted according to mass ratios of each constituent in the daily inflows. For example, if Santa Ana Delhi had an NH3 concentration of 1 mg/L and flow rate of 1 m³/s, and San Diego creek had an NH3 concentration of 5 mg/L and flow rate of 2 m³/s, the mass ratios of NH3 of Santa Ana Delhi and San Diego Creek would be 1 mg/L × 1 m³/s / (1 mg/L × 1 m³/s + 5 mg/L × 2 m³/s) and 5 mg/L × 2 m³/s / (1 mg/L × 1 m³/s + 5 mg/L × 2 m³/s), respectively. A reduction in the ratio of particulate nitrogen to suspended sediment corresponding to the storm flow limit was applied as well, assuming a reduction in the nitrogen content of sediment under reduced loading alternatives. The ratio of particulate nitrogen to suspended sediment used in the calibration/validation simulations was 0.001. For the simulations with storm flows limited to 3 mg/L total nitrogen, the particulate nitrogen to suspended sediment ratio was set to .0003 assuming a 70% reduction of particulate nitrogen in the suspended sediments coming from inflows. For the simulations with storm flows limited to 5 mg/L total nitrogen, the ratio was set to .0005, assuming a 50% reduction of particulate nitrogen in the suspended sediments coming from inflows. Concentration limits and particulate nitrogen to suspended sediment ratios are summarized for each alternative in Table 7-1.

Simulations were performed for the 2004 through 2006 period and the model calibration/validation results are used as the Base case for comparison. Figure 7-1 through Figure 7-6 show time series of total nitrogen entering the Bay in the inflows (San Diego Creek and Santa Ana Delhi Channel) for each of the alternatives in comparison with the Base case. Annually averaged total nitrogen is plotted in Figure 7-7 and Figure 7-8 for each of the inflows and annual median total nitrogen is plotted in Figure 7-9 and Figure 7-10 for each of the inflows. Note that annual medians are nearly identical for the alternatives with the same low flow loading (e.g. Alternative 1 and Alternative 4) and thus they plot on top of each other.
Alternative simulation results are plotted with the Base case for 2004 in Figure 7-11 through Figure 7-19, 2005 in Figure 7-20 through Figure 7-28 and 2006 in Figure 7-29 through Figure 7-37. The first plot for each year shows time series total macroalgae mass in the Bay and the next plots show time series of macroalgae concentration at each of the algae sampling locations. Alternatives 1 – 3, those with the greatest reduction in storm flow nitrogen load, produce the lowest concentrations of macroalgae. Alternative 1, with the lowest total nitrogen limits for both storm and low flow periods, reduces the peak total macroalgae mass by just over 50% for each year. Alternative 6, with the highest total nitrogen limits, reduces the peak total macroalgae mass by approximately 30% for each year.

Table 7-1  Summary of concentration limits and particulate nitrogen to suspended sediment ratios for alternatives.

<table>
<thead>
<tr>
<th></th>
<th>Total Nitrogen Concentration limit (mg/L)</th>
<th>Particulate nitrogen to suspended sediment ratio</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>Storm flow (&gt;50 cfs)</td>
<td>Low flow (≤50 cfs)</td>
</tr>
<tr>
<td>Base case</td>
<td>No limit</td>
<td>No limit</td>
</tr>
<tr>
<td>Alternative 1</td>
<td>3</td>
<td>0.5</td>
</tr>
<tr>
<td>Alternative 2</td>
<td>3</td>
<td>1</td>
</tr>
<tr>
<td>Alternative 3</td>
<td>3</td>
<td>2</td>
</tr>
<tr>
<td>Alternative 4</td>
<td>5</td>
<td>0.5</td>
</tr>
<tr>
<td>Alternative 5</td>
<td>5</td>
<td>1</td>
</tr>
<tr>
<td>Alternative 6</td>
<td>5</td>
<td>2</td>
</tr>
</tbody>
</table>
Figure 7-1 Time series of total nitrogen entering the Bay from Santa Ana Delhi and San Diego Creek for Base case and Alternative 1 (3.0 mg/L-0.5 mg/L).

Figure 7-2 Time series of total nitrogen entering the Bay from Santa Ana Delhi and San Diego Creek for Base case and Alternative 2 (3.0 mg/L-1.0 mg/L).
Figure 7-3 Time series of total nitrogen entering the Bay from Santa Ana Delhi and San Diego Creek for Base case and Alternative 3 (3.0 mg/L-2.0 mg/L).

Figure 7-4 Time series of total nitrogen entering the Bay from Santa Ana Delhi and San Diego Creek for Base case and Alternative 4 (5.0 mg/L-0.5 mg/L).
Figure 7-5 Time series of total nitrogen entering the Bay from Santa Ana Delhi and San Diego Creek for Base case and Alternative 5 (5.0 mg/L-1.0 mg/L).

Figure 7-6 Time series of total nitrogen entering the Bay from Santa Ana Delhi and San Diego Creek for Base case and Alternative 6 (5.0 mg/L-2.0 mg/L).
Figure 7-7 Annually averaged total nitrogen concentration of Santa Ana Delhi for Base case and alternatives.

Figure 7-8 Annually averaged total nitrogen concentration of San Diego Creek for Base case and alternatives.
Figure 7-9 Annual median of total nitrogen concentration of Santa Ana Delhi for Base case and alternatives. Note that the alternatives with the same dry season loading (e.g. 1 and 4) have nearly identical annual medians and thus plot on top of one another.

Figure 7-10 Annual median of total nitrogen concentration of San Diego Creek for Base case and alternatives. Note that the alternatives with the same dry season loading (e.g. 1 and 4) have nearly identical annual medians and thus plot on top of one another.
Figure 7-11 Comparison of total dry mass of macroalgae among alternatives with the Base case in 2004.
Figure 7-12 Comparison of macroalgae concentrations among alternatives with the Base case at station ALG24 in 2004.

Figure 7-13 Comparison of macroalgae concentrations among alternatives with the Base case at station ALG19 in 2004.
Figure 7-14 Comparison of macroalgae concentrations among alternatives with the Base case at station ALG16 in 2004.

Figure 7-15 Comparison of macroalgae concentrations among alternatives with the Base case at station ALG13 in 2004.
Figure 7-16 Comparison of macroalgae concentrations among alternatives with the Base case at station ALG9 in 2004.

Figure 7-17 Comparison of macroalgae concentrations among alternatives with the Base case at station ALG7 in 2004.
Figure 7-18 Comparison of macroalgae concentrations among alternatives with the Base case at station ALG4 in 2004.

Figure 7-19 Comparison of macroalgae concentrations among alternatives with the Base case at station ALG2 in 2004.
Figure 7-20 Comparison of total dry mass of macroalgae among alternatives with the Base case in 2005.
Figure 7-21 Comparison of macroalgae concentrations among alternatives with the Base case at station ALG24 in 2005.

Figure 7-22 Comparison of macroalgae concentrations among alternatives with the Base case at station ALG19 in 2005.
Figure 7-23 Comparison of macroalgae concentrations among alternatives with the Base case at station ALG16 in 2005.

Figure 7-24 Comparison of macroalgae concentrations among alternatives with the Base case at station ALG13 in 2005.
Figure 7-25 Comparison of macroalgae concentrations among alternatives with the Base case at station ALG9 in 2005.

Figure 7-26 Comparison of macroalgae concentrations among alternatives with the Base case at station ALG7 in 2005.
Figure 7-27 Comparison of macroalgae concentrations among alternatives with the Base case at station ALG4 in 2005.

Figure 7-28 Comparison of macroalgae concentrations among alternatives with the Base case at station ALG2 in 2005.
Figure 7-29 Comparison of total dry mass of macroalgae among alternatives with the Base case in 2006.
Figure 7-30 Comparison of macroalgae concentrations among alternatives with the Base case at station ALG24 in 2006.

Figure 7-31 Comparison of macroalgae concentrations among alternatives with the Base case at station ALG19 in 2006.
Figure 7-32 Comparison of macroalgae concentrations among alternatives with the Base case at station ALG16 in 2006.

Figure 7-33 Comparison of macroalgae concentrations among alternatives with the Base case at station ALG13 in 2006.
Figure 7-34 Comparison of macroalgae concentrations among alternatives with the Base case at station ALG9 in 2006.

Figure 7-35 Comparison of macroalgae concentrations among alternatives with the Base case at station ALG7 in 2006.
Figure 7-36 Comparison of macroalgae concentrations among alternatives with the Base case at station ALG4 in 2006.

Figure 7-37 Comparison of macroalgae concentrations among alternatives with the Base case at station ALG2 in 2006.
8. Summary and conclusions

The Newport Bay water quality model has been updated and calibrated to support short term and long term management of Newport Bay and aid the Regional Board in determination of new numeric water quality objectives for the TMDL implementation plan.

Previous versions of the model have simulated the occurrence of macroalgal blooms, however, the model could not predict yearly variations in the magnitude of the blooms. In this study, the water quality model for Newport Bay has been updated with the sediment nutrient fluxes based on the SCCWRP sediment study, a review of relevant literature and available data. The model was calibrated and validated using new data sets for 2004 to 2006 from the annual monitoring program performed by the Regional Board.

The key improvement to the model is the addition of a sediment flux model. Sediment transport simulations were performed to provide appropriate sediment depth and distribution to the sediment flux model.

The model was successfully calibrated and validated over the three-year period of 2004 through 2006 to assure that variations in the magnitude of annual macroalgae blooms resulting from different year types was reproduced by the model. There was a particularly large sediment load entering the Bay in 2005, with corresponding large summer macroalgae blooms. The model was able to reproduce this larger bloom relative to years 2004 and 2006.

Overall, the model calibration and validation results were in reasonable agreement with macroalgae, nutrient, dissolved oxygen, temperature and porewater nutrient observations.

An examination of limitation factors from the model indicates that nitrogen and phosphorus are the most important factors controlling macroalgae blooms in Newport Bay.

The calibrated model was run in a predictive mode to evaluate the impacts of alternative total nitrogen load levels entering the Bay. Six alternative nitrogen loading levels were analyzed with different total nitrogen limits during storm and low flow periods. The fraction of particulate nitrogen in the suspended sediment was set to reduced levels corresponding to the storm flow nitrogen limits. Alternative simulation results indicate that reductions in total nitrogen in storm flows and fraction of particulate nitrogen in the sediment have a greater impact on reducing macroalgae blooms than reduction in total nitrogen inputs during low flow periods.

Simulations were performed to compare results from the calibrated model to results from simulations with no sediment interactions. Elimination of the sediment interactions reduced
macroalgae biomass to minimal amounts, thus confirming the importance of sediments as a nutrient source for macroalgae blooms.

For future study, development of a sediment flux model which better represents an anaerobic environment or a sediment model with resuspension induced by waves and currents in the bay may improve model results.
9. References

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